

# REPORT DOCUMENTATION PAGE

*Form Approved  
OMB No. 0704-0188*

Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing this collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden to Department of Defense, Washington Headquarters Services, Directorate for Information Operations and Reports (0704-0188), 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302. Respondents should be aware that notwithstanding any other provision of law, no person shall be subject to any penalty for failing to comply with a collection of information if it does not display a currently valid OMB control number. **PLEASE DO NOT RETURN YOUR FORM TO THE ABOVE ADDRESS.**

<b>1. REPORT DATE (DD-MM-YYYY)</b> July 2015	<b>2. REPORT TYPE</b> Briefing Charts	<b>3. DATES COVERED (From - To)</b> July 2015-August 2015		
<b>4. TITLE AND SUBTITLE</b> IWetting properties of polysiloxane networks modified in situ with fluoroalkyl-substituted linear and POSS cage structures (Briefing Charts)			<b>5a. CONTRACT NUMBER</b> In-House	
			<b>5b. GRANT NUMBER</b>	
			<b>5c. PROGRAM ELEMENT NUMBER</b>	
<b>6. AUTHOR(S)</b> Raymond Campos, Sean Ramirez, Joseph Mabry			<b>5d. PROJECT NUMBER</b>	
			<b>5e. TASK NUMBER</b>	
			<b>5f. WORK UNIT NUMBER</b> Q0BG	
<b>7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES)</b> Air Force Research Laboratory (AFMC) AFRL/RQRP 10 E. Saturn Blvd. Edwards AFB, CA93524-7680			<b>8. PERFORMING ORGANIZATION REPORT NO.</b>	
<b>9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES)</b> Air Force Research Laboratory (AFMC) AFRL/RQR 5 Pollux Drive Edwards AFB CA 93524-7048			<b>10. SPONSOR/MONITOR'S ACRONYM(S)</b>	
			<b>11. SPONSOR/MONITOR'S REPORT NUMBER(S)</b> <b>AFRL-RQ-ED-VG-2015-291</b>	
<b>12. DISTRIBUTION / AVAILABILITY STATEMENT</b> Distribution A: Approved for Public Release; Distribution Unlimited.				
<b>13. SUPPLEMENTARY NOTES</b> Briefing Charts presented at 250th ACS National Meeting; Boston, MA; 16-20 August 2015. PA#15439.				
<b>14. ABSTRACT</b> Briefing Charts				
<b>15. SUBJECT TERMS</b>				
<b>16. SECURITY CLASSIFICATION OF:</b>		<b>17. LIMITATION OF ABSTRACT</b> SAR	<b>18. NUMBER OF PAGES</b> 38	<b>19a. NAME OF RESPONSIBLE PERSON</b> Joseph Mabry
<b>a. REPORT</b> Unclassified	<b>b. ABSTRACT</b> Unclassified	<b>c. THIS PAGE</b> Unclassified	<b>19b. TELEPHONE NO</b> <i>(include area code)</i> 661-275-5857	

# Wetting properties of polysiloxane networks modified in situ with fluoroalkyl-substituted linear and POSS cage structures

Raymond Campos, Sean M. Ramirez,  
and Joesph M. Mabry

August 17, 2015

ACS National Conference

# Paradigms for low energy surfaces

## Crystalline surfaces

- Immobilized chemical moieties with low polarizability (e.g. fluorine)  
 $\text{CF} > \text{CF}_2 > \text{CF}_3 > \text{SF}_5$
- High crystallinity to prevent surface reorganization when in contact with liquids, biofouling sources, etc.

Extreme liquid repellency when combined with surface roughness and re-entrant geometry

## “liquid-like” surfaces

- low energy barriers between metastable states
- Metal oxide surfaces and liquid-infused materials (e.g. SLIPS)

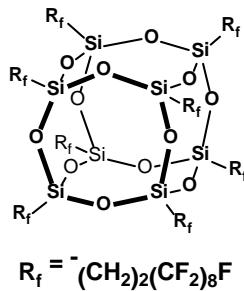
Fluorine content not required to repel low surface tension liquids?

Low CA hysteresis despite lower contact angle values

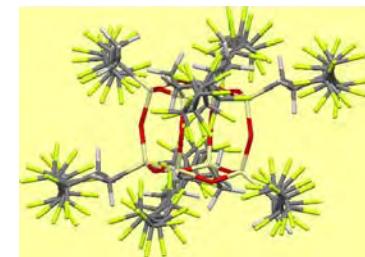
# Surface modification of PDMS/siloxane networks

- Primarily oxide formation via O<sub>2</sub> plasma, UV/ozone, etc. and subsequent functionalization
- Functional PDMS (e.g. residual vinyl groups in network)
- Functional silsesquioxane networks

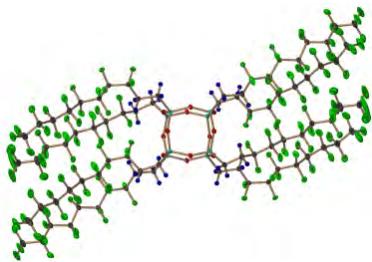
# Properties of Fluorodecyl<sub>8</sub>T<sub>8</sub> POSS



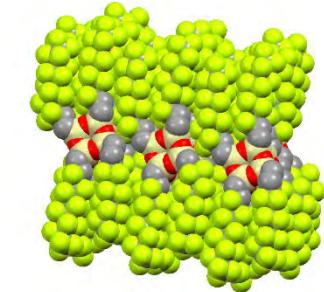
Extremely low surface energy



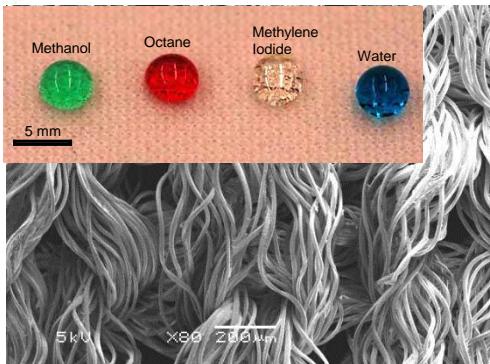
Surface migration in polymers



Surface responsive behavior

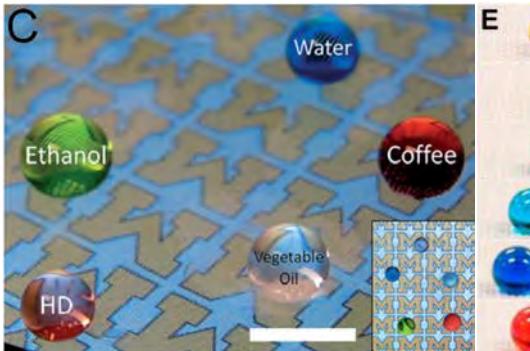


Enabling....



**Superomniphobic fabrics via dip-coating**

Choi *et al.*, Ang. Chem., 2009

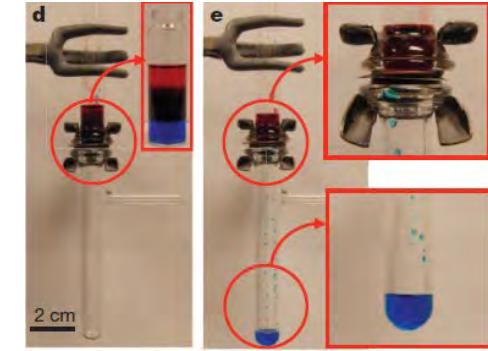


**Transparent  
Omniphobicity**

Golovin *et al.*, Ang. Chem., 2013

**Extreme  
Omniphobicity**

Pan *et al.*, JACS., 2012



**Oil/water emulsion gravity seperation**

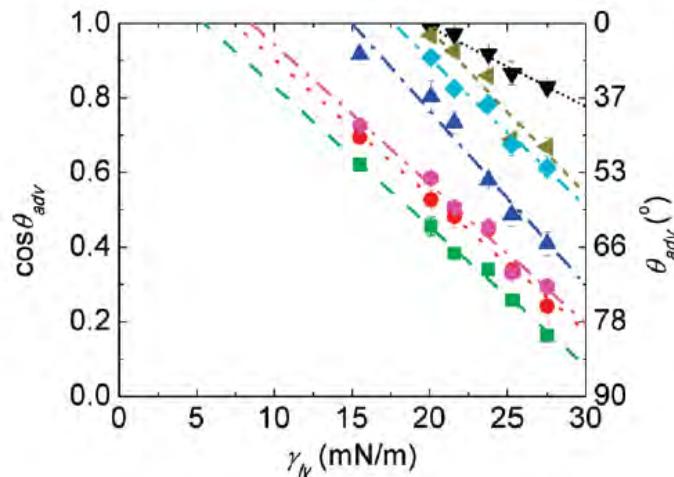
Kota *et al.*, JACS., 2012

# Surface Energy of Fluorodecyl<sub>8</sub>T<sub>8</sub> POSS

fluorodecyl<sub>8</sub>T<sub>8</sub> POSS

5.5 mN/m (Zisman analysis)\*

8.8 – 10.2 mN/m (Girafalco-Good analysis)\*



Zisman analysis  
(only alkane probing liquids used)

	Fluorodecyl $T_8$ , R = $-(CH_2)_2-(CF_2)_7-CF_3$
	Fluoroctyl $T_8$ , R = $-(CH_2)_2-(CF_2)_5-CF_3$
	Fluorohexyl $T_8$ , R = $-(CH_2)_2-(CF_2)_3-CF_3$
	Fluoropropyl $T_8$ , R = $-(CH_2)_2-CF_3$
	Hexafluoro-i-butyl $T_8$ , R = $-CH_2-CH(CF_3)_2$
	Fluorodecyl $Q_4$ , R = $-(CH_2)_2-(CF_2)_7-CF_3$
	Fluorodecyl M2, R = $-(CH_2)_2-(CF_2)_7-CF_3$

\*Chhatre, S. S.; Guardado, J. O.; Moore, B. M.; Haddad, T. S.; Mabry, J. M.; McKinley, G. H.; Cohen, R. E., Fluoroalkylated Silicon-Containing Surfaces-Estimation of Solid-Surface Energy. *ACS Applied Materials and Interfaces* 2010, 2 (12), 3544-3554.

# Surface Energy of Fluorodecyl<sub>8</sub>T<sub>8</sub> POSS

fluorodecyl<sub>8</sub>T<sub>8</sub> POSS

5.5 mN/m (Zisman analysis)\*  
8.8 - 10.2 mN/m (Girafalco-Good analysis)\*

Why?

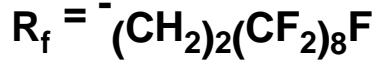
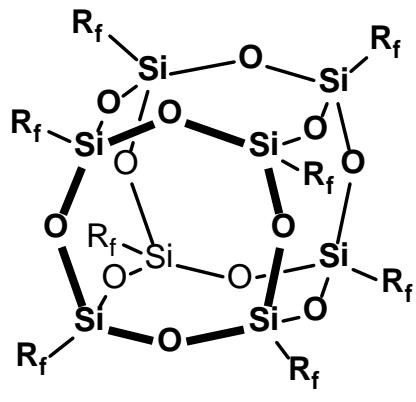
Polytetrafluoroethylene  
 $\text{CF}_3-(\text{CF}_2)_n-\text{CF}_3$

18 - 20 mN/m (Zisman analysis)

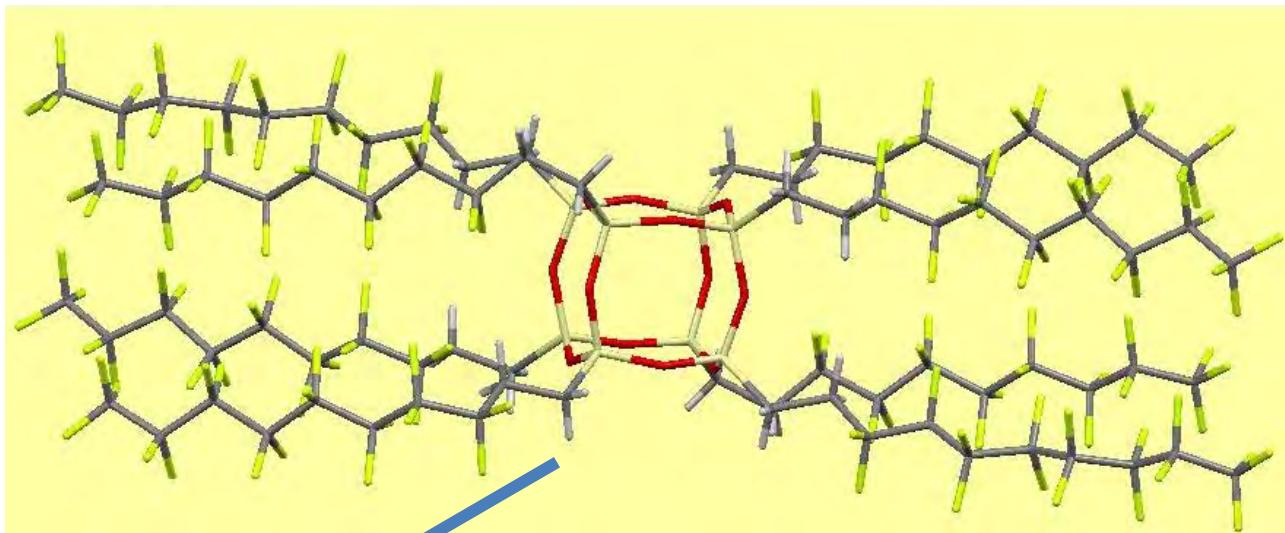
CF<sub>3</sub> monolayer

6.7 mN/m (Zisman analysis)

\*Chhatre, S. S.; Guardado, J. O.; Moore, B. M.; Haddad, T. S.; Mabry, J. M.; McKinley, G. H.; Cohen, R. E., Fluoroalkylated Silicon-Containing Surfaces-Estimation of Solid-Surface Energy. *ACS Applied Materials and Interfaces* 2010, 2 (12), 3544-3554.



Fluorodecyl<sub>8</sub>T<sub>8</sub> POSS



Rotate ~ 90°

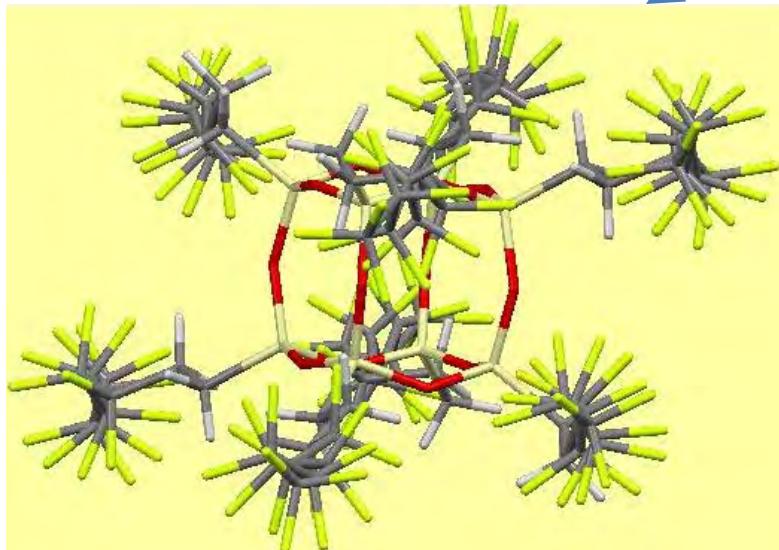
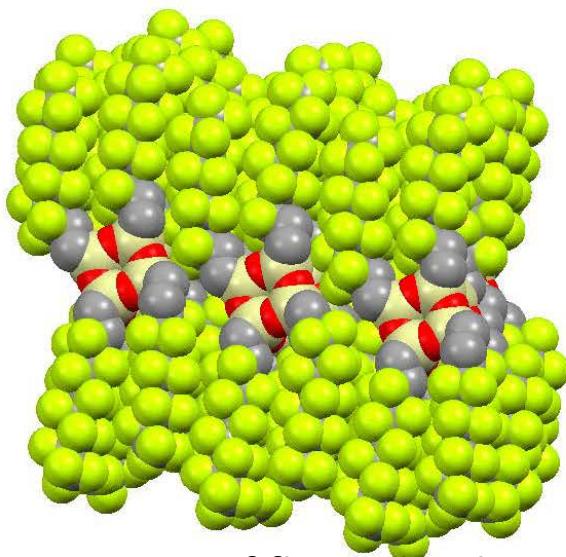
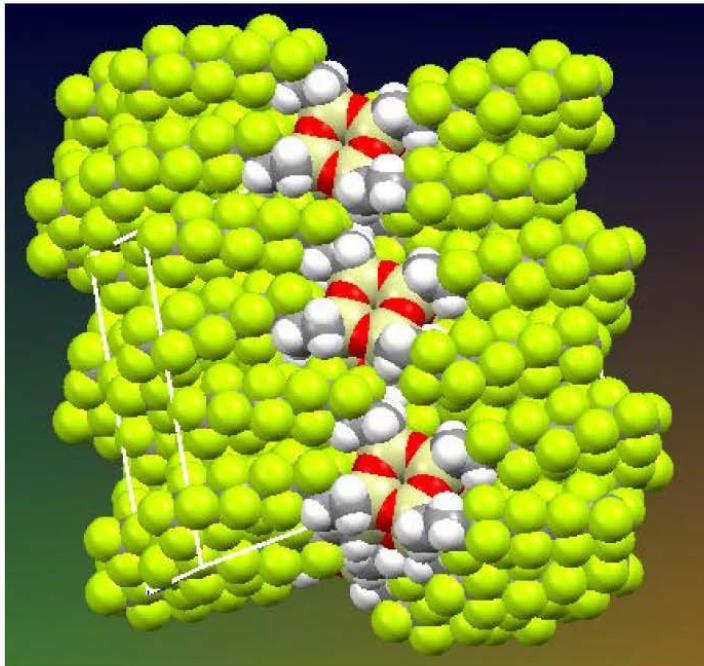
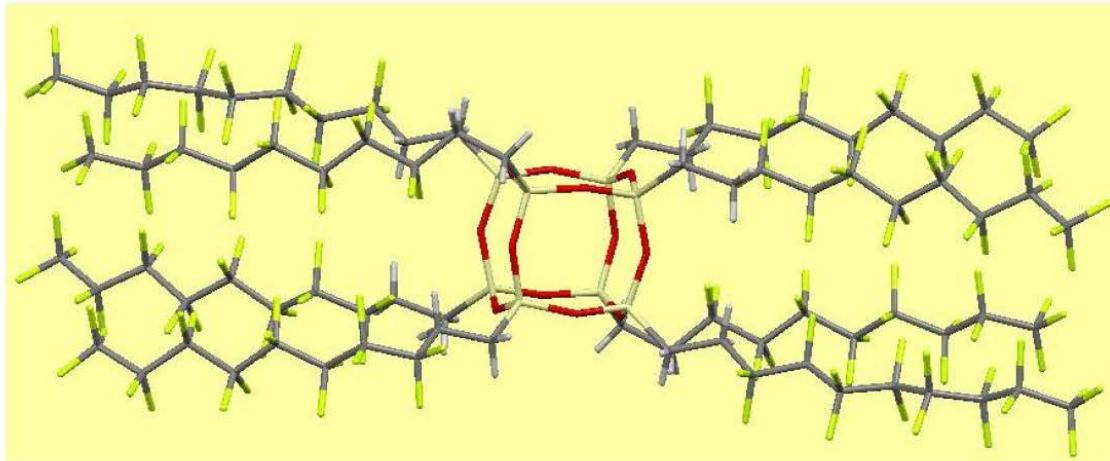
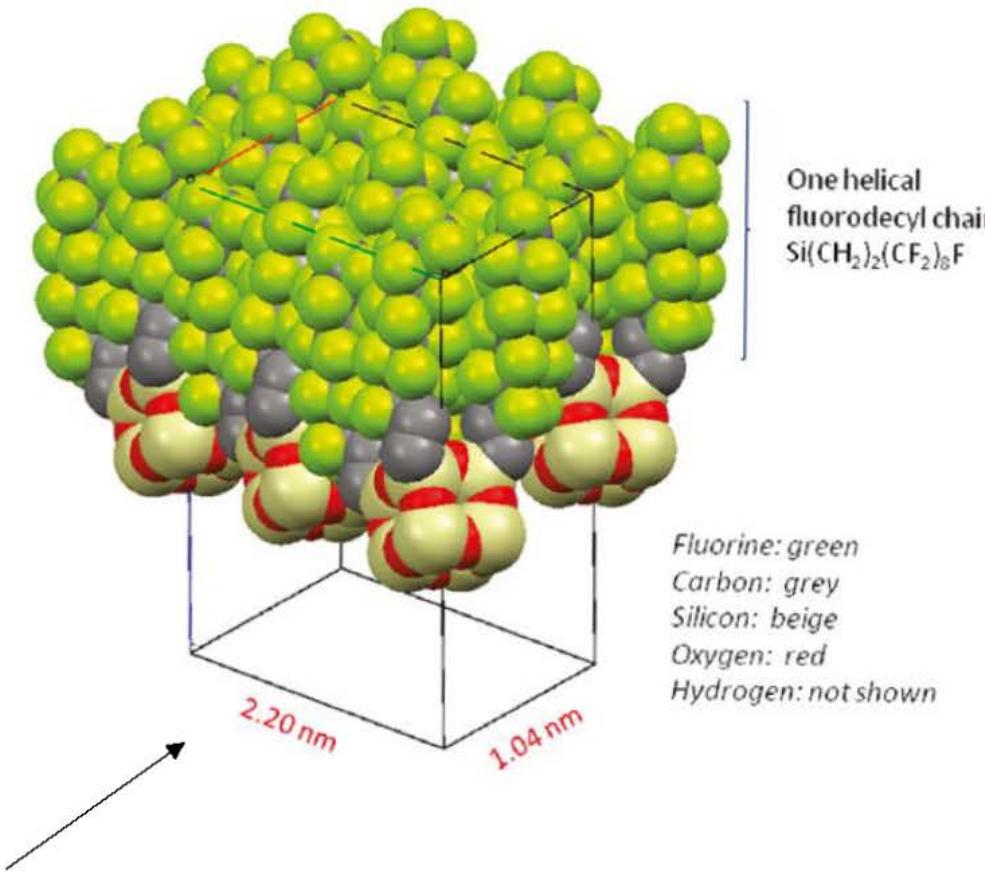
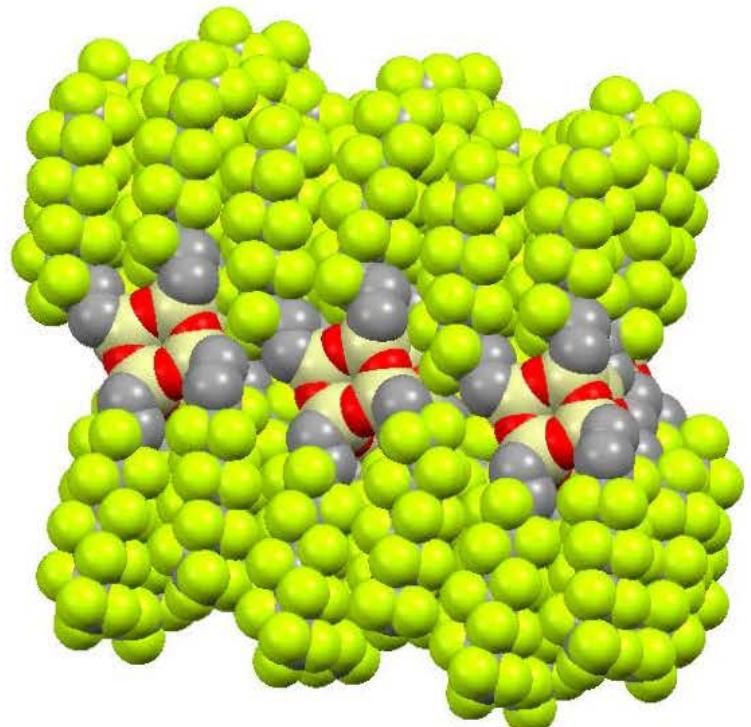


Image displaying the helical conformation of fluorodecyl substituents in the solid state packing of fluorodecyl<sub>8</sub>T<sub>8</sub> POSS

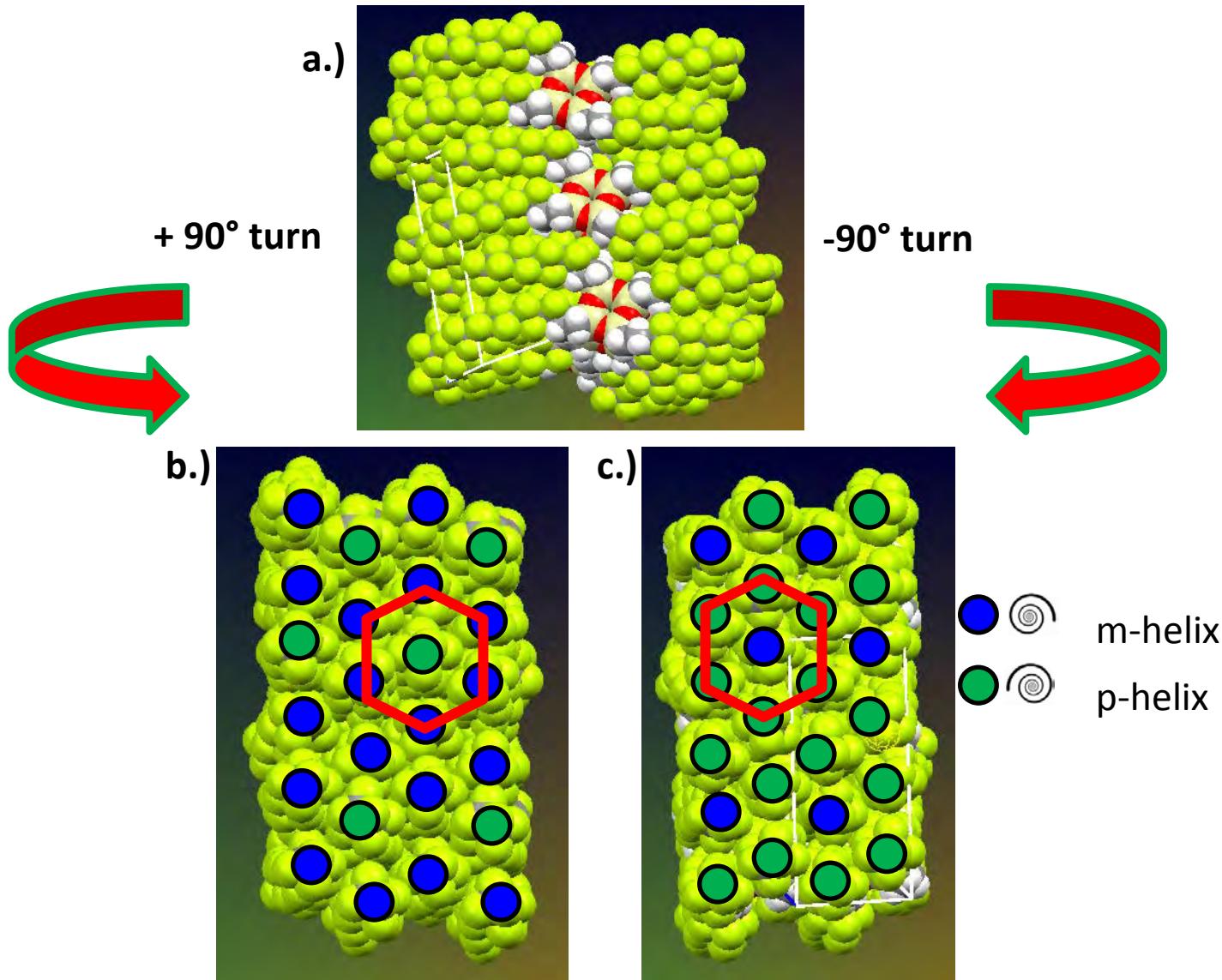


3 x 2 array of fluorodecyl8T8 POSS cages displaying a lamella-type packing

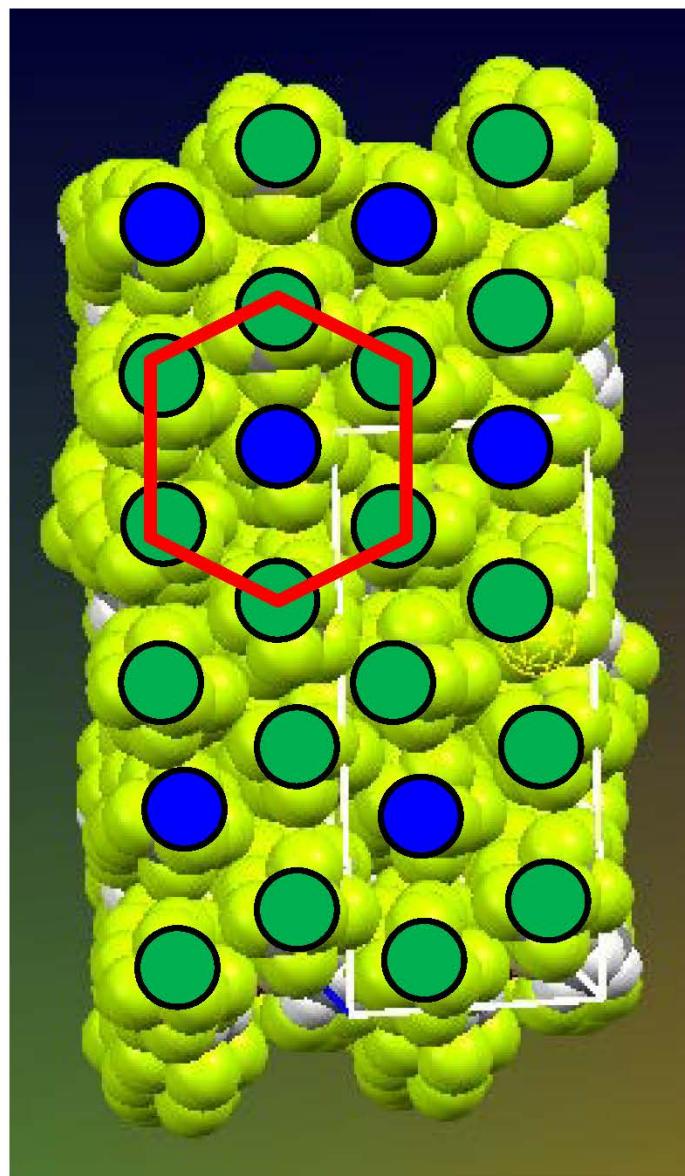
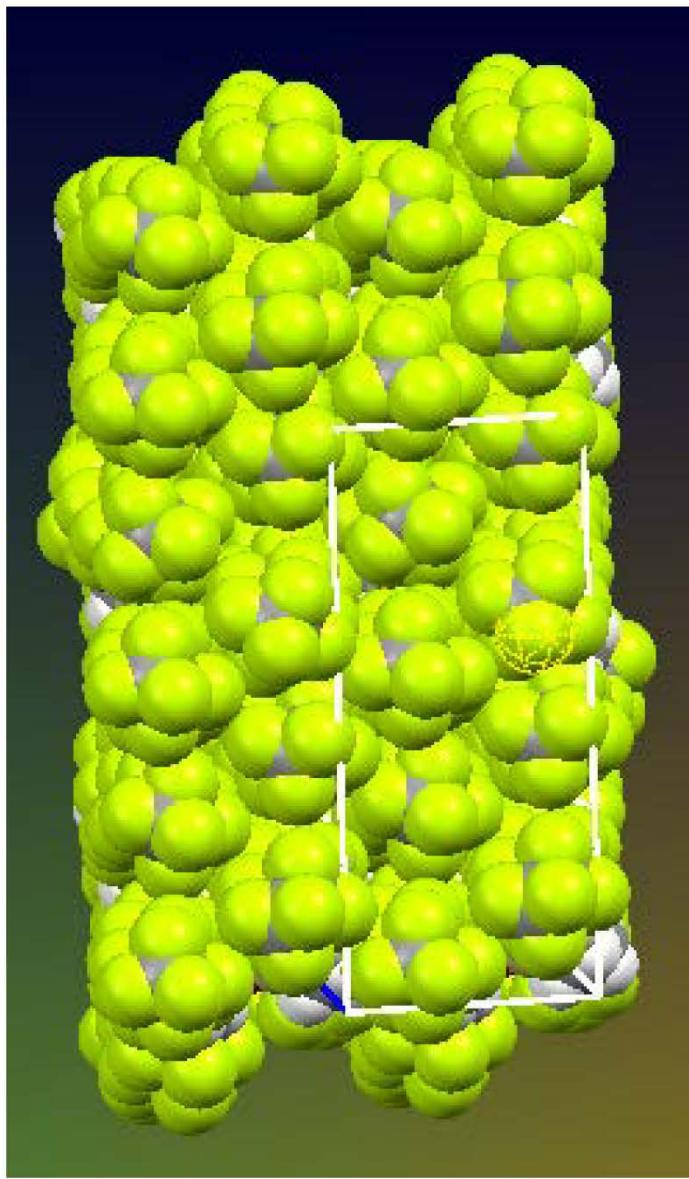
Distribution A: Approved for public release; distribution is unlimited.



Campos, R., Haddad T. S., et al., *Langmuir* **2011**, *27* (16), 10206-10215.



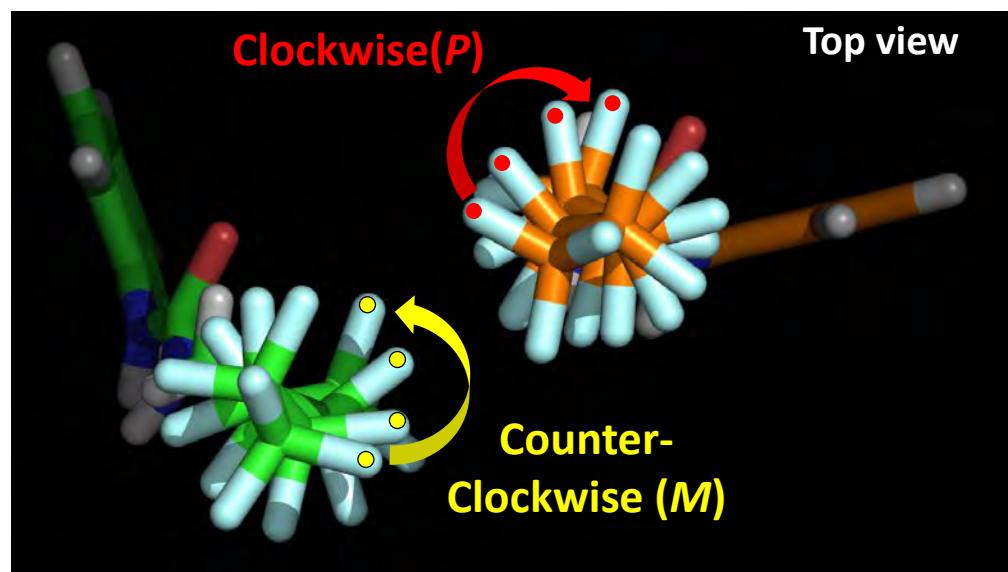
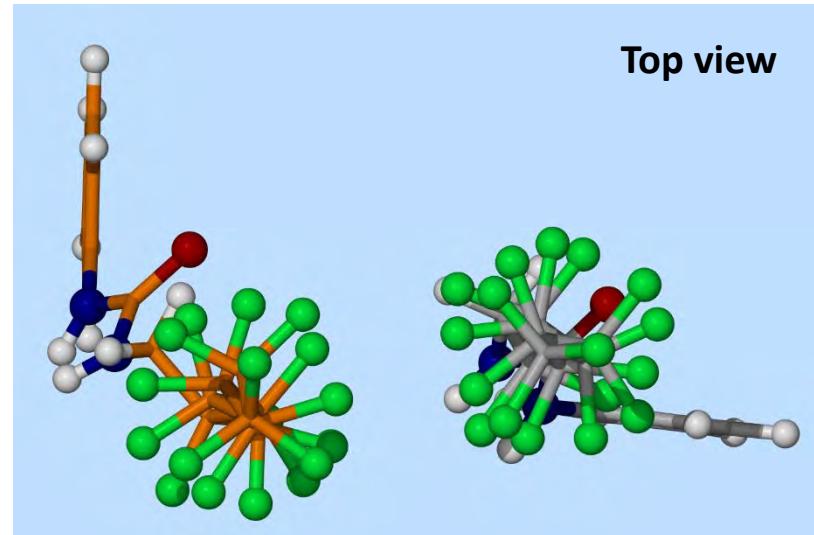
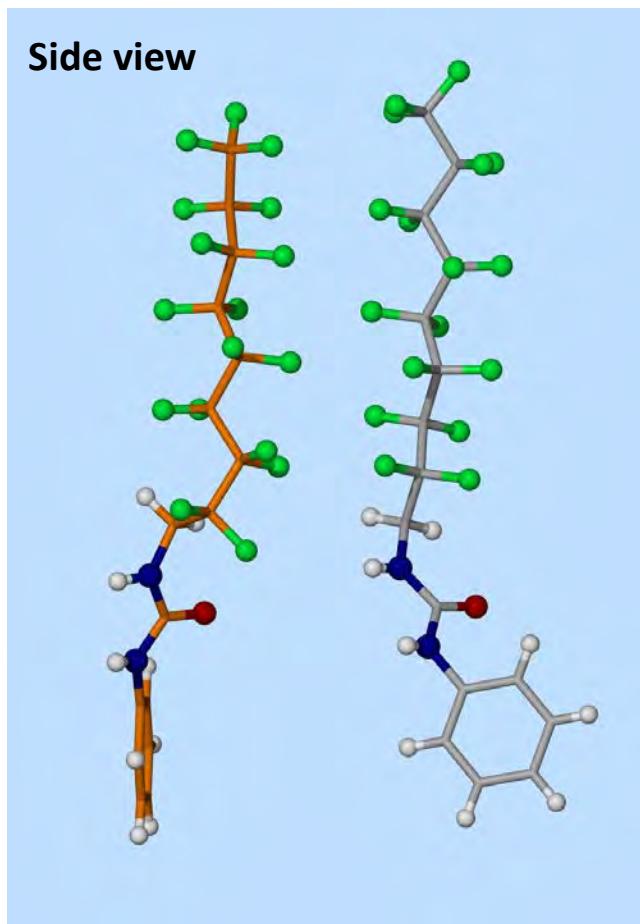
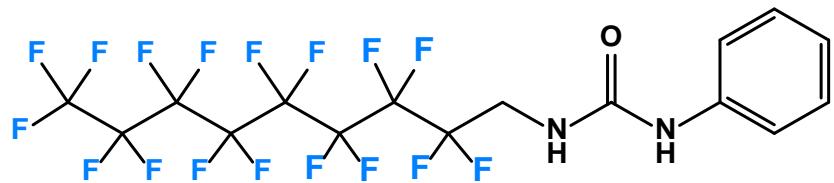
Distribution A: Approved for public release; distribution is unlimited.



m-helix  
 p-helix

Distribution A: Approved for public release; distribution is unlimited.

# Helical crystal packing of fluoroalkyl-substituted urea

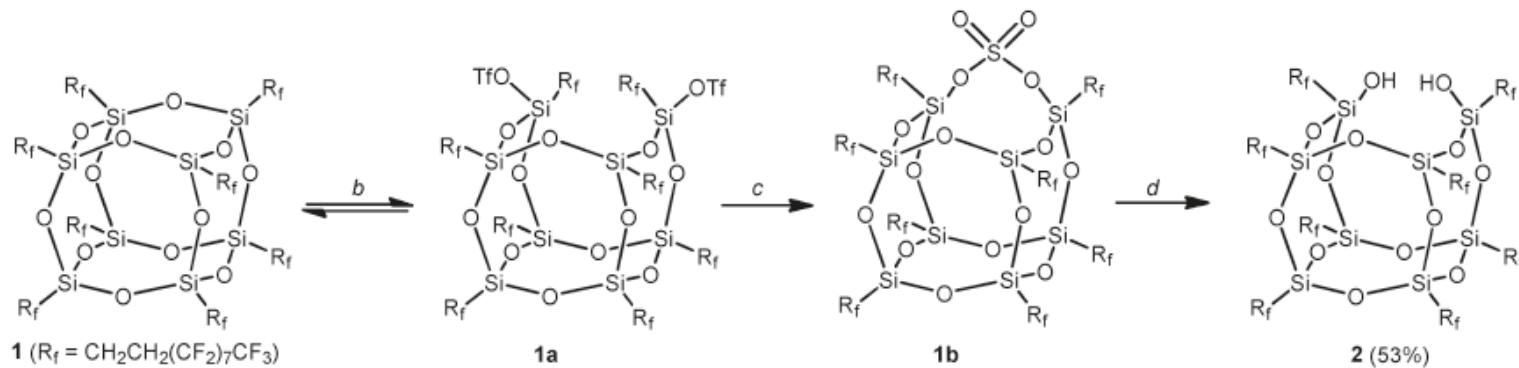


# Helical crystal packing of fluoroalkyl-substituted urea



# Functional Fluorodecyl POSS Compounds Enabled by Incompletely Condensed Intermediate

Scheme 1. Synthesis of Incompletely Condensed Fluoroalkyl Silsesquioxane<sup>a</sup>

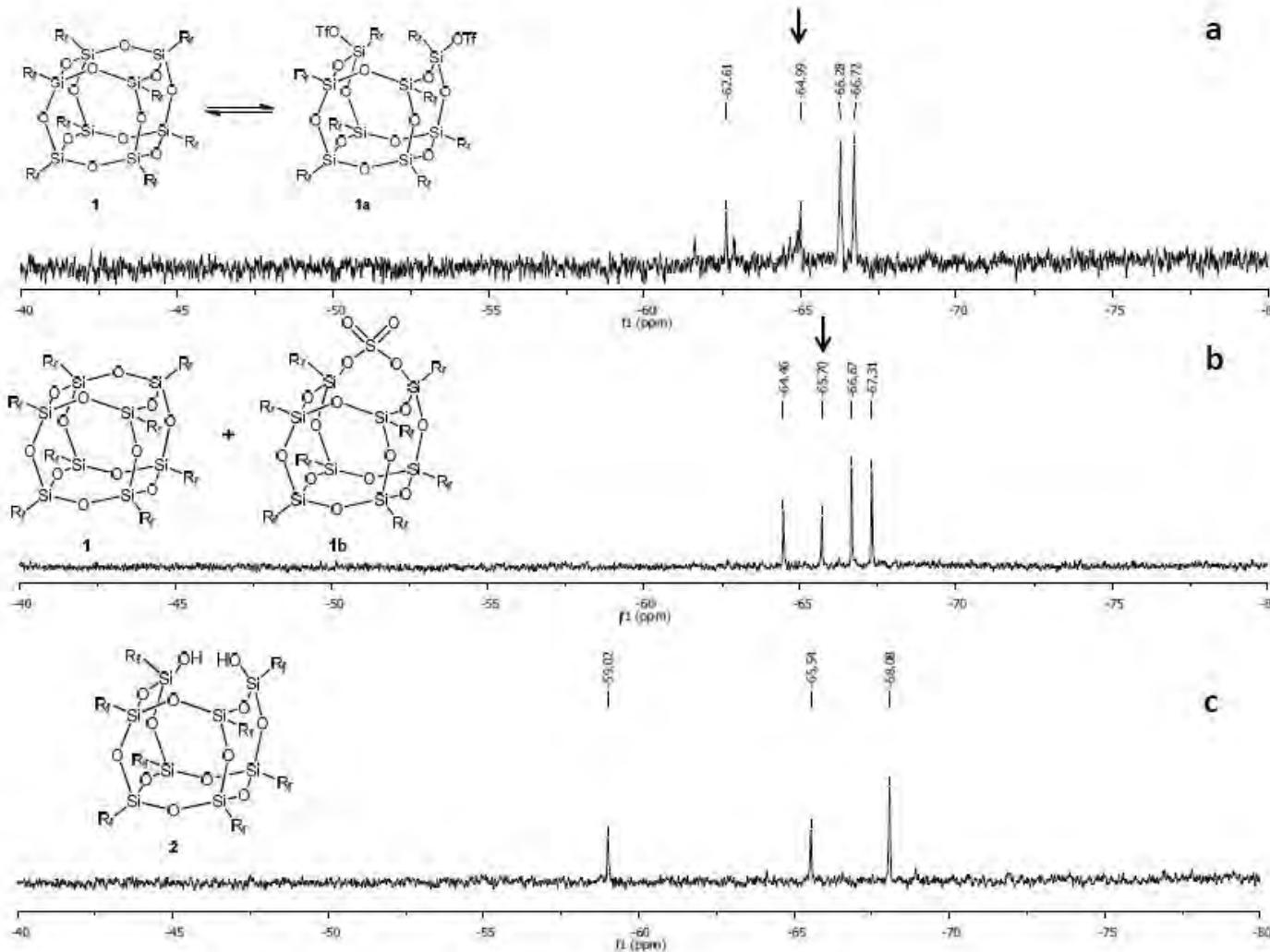


<sup>a</sup> Conditions: All reactions were performed in  $C_6F_6$  at 25 °C. <sup>b</sup> $CF_3SO_3H$ , 75 min; <sup>c</sup> $NBu_4HSO_4$ , 30 min; <sup>d</sup> $(CF_3)_2CH_2OH/H_2O$  (10:1), 12 h.

Ramirez, S. M., Diaz, Y. J., Campos, R., Stone, R. L., Haddad, T. S., Mabry, J. M. *JACS*, **2011**, 133, 20084.

Distribution A: Approved for public release; distribution is unlimited.

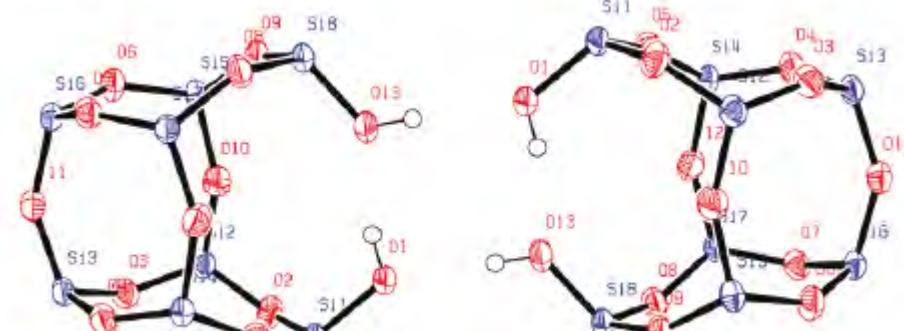
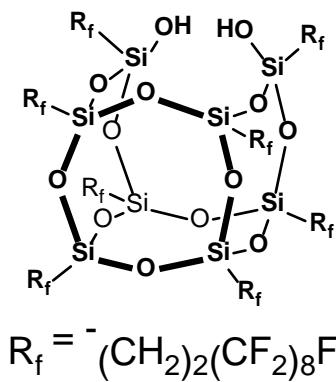
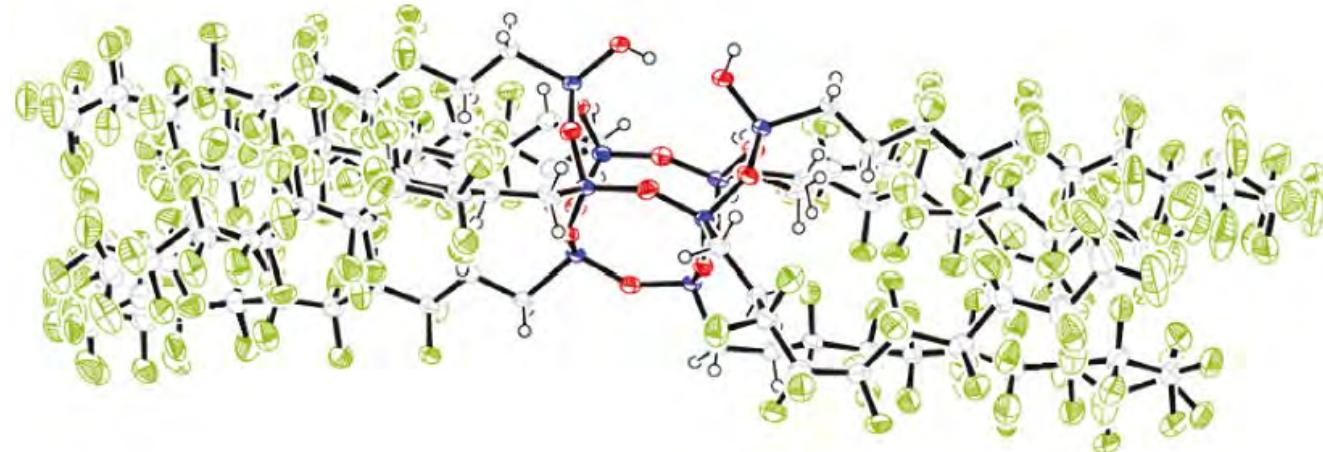
# $^{29}\text{Si}$ NMR Spectra of fluorodecyl POSS disilanol Intermediates and Product



Ramirez, S. M., Diaz, Y. J., Campos, R., Stone, R. L., Haddad, T. S., Mabry, J. M. *JACS*, 2011, 133, 20084.

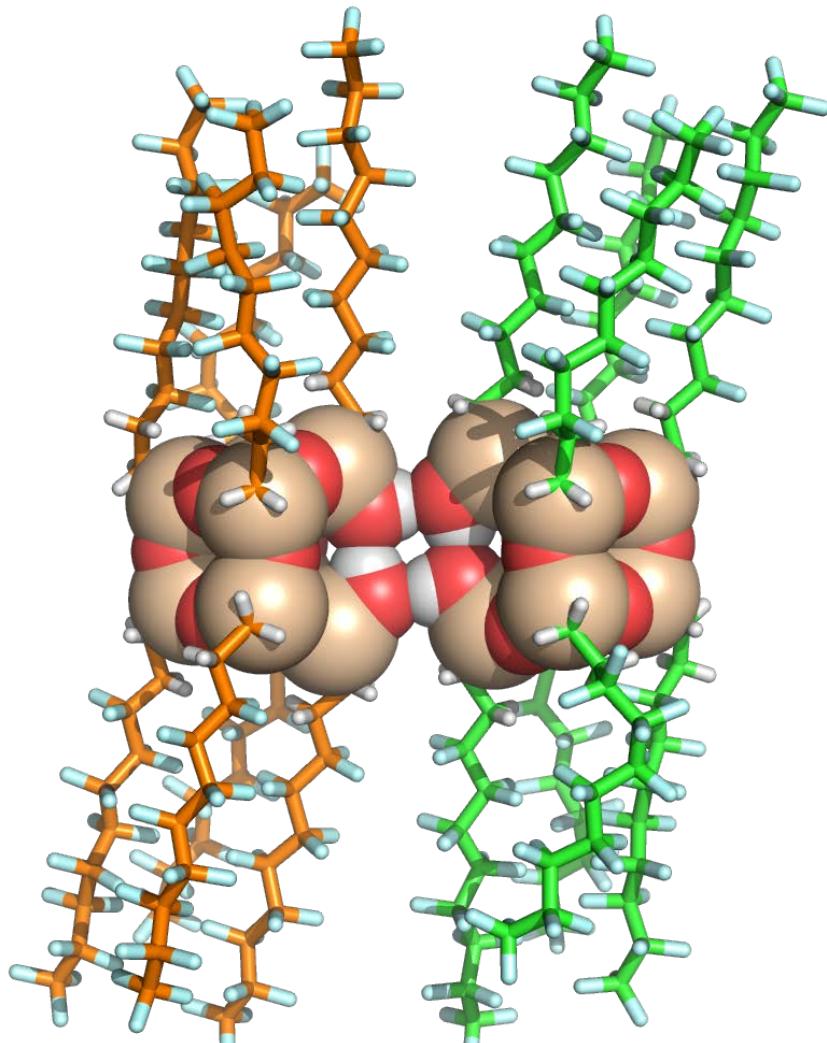
Distribution A: Approved for public release; distribution is unlimited.

# ORTEP representations of Fluorodecyl POSS Disilanol crystal structure

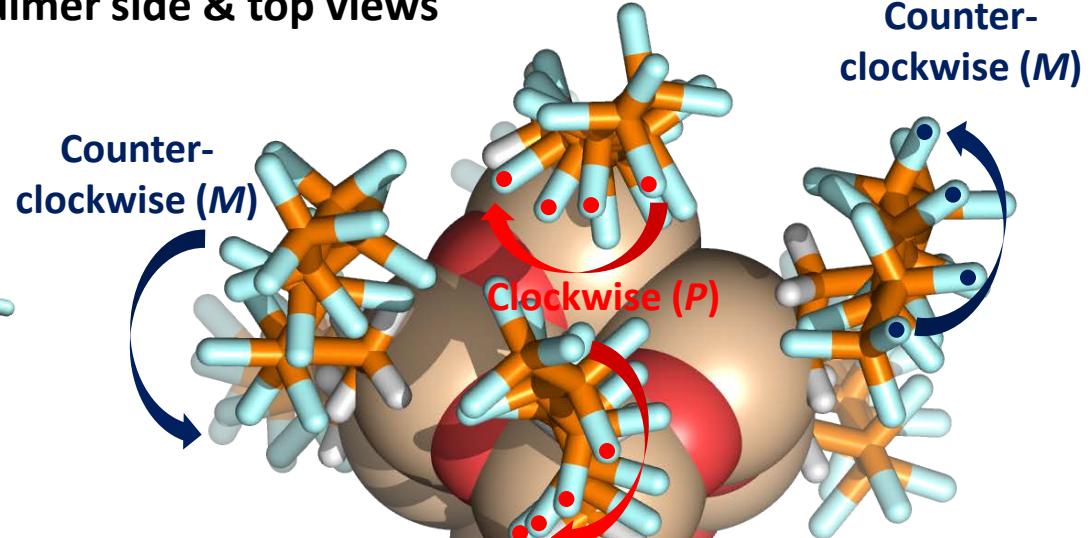


Ramirez, S. M., Diaz, Y. J., Campos, R., Stone, R. L., Haddad, T. S., Mabry, J. M. *JACS*, 2011, 133, 20084.

## F-POSS cage dimer side & top views



Side view



Counter-clockwise (M)

Counter-clockwise (M)

Clockwise (P)

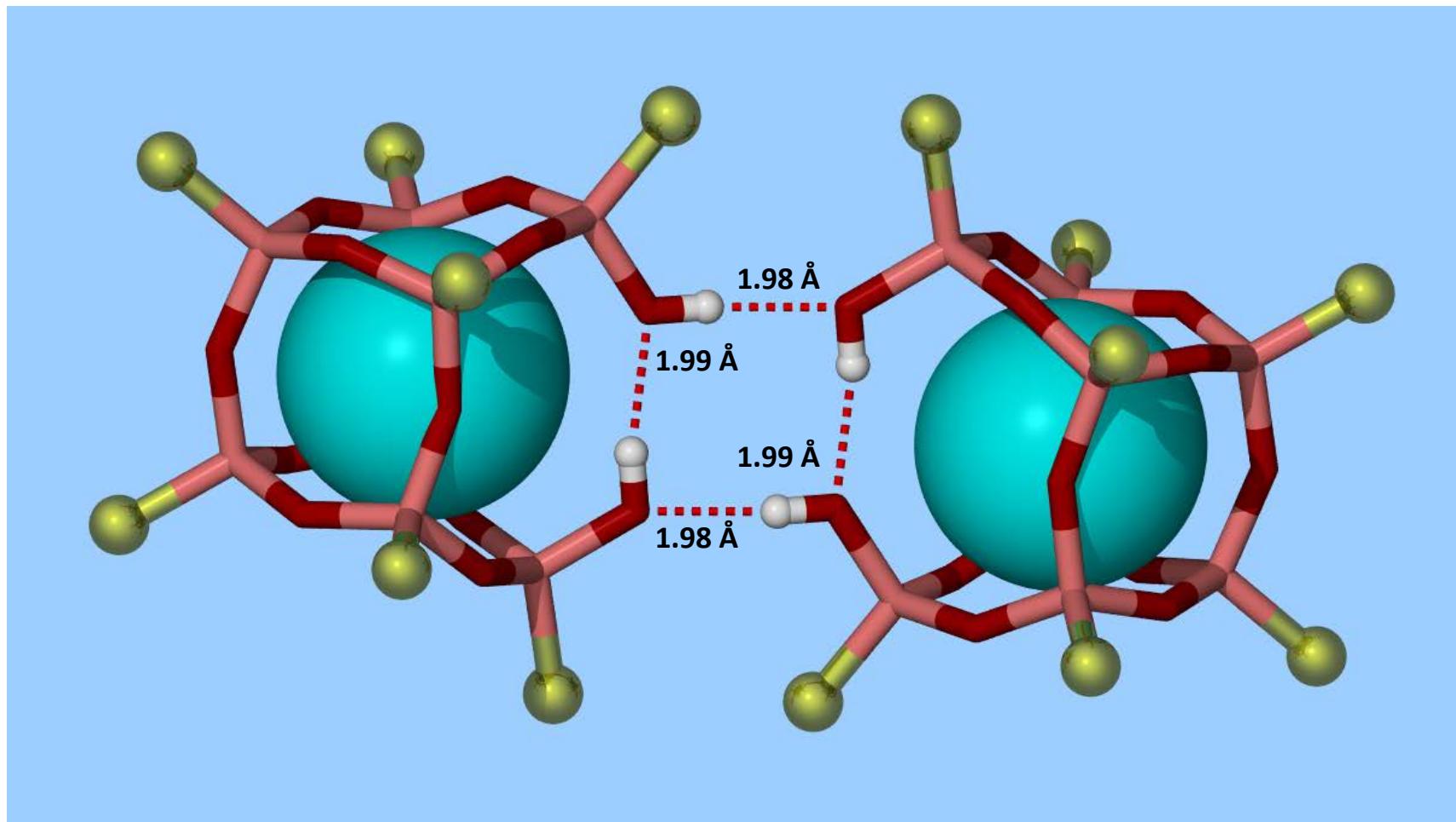
Clockwise (P)

Counter-clockwise (M)

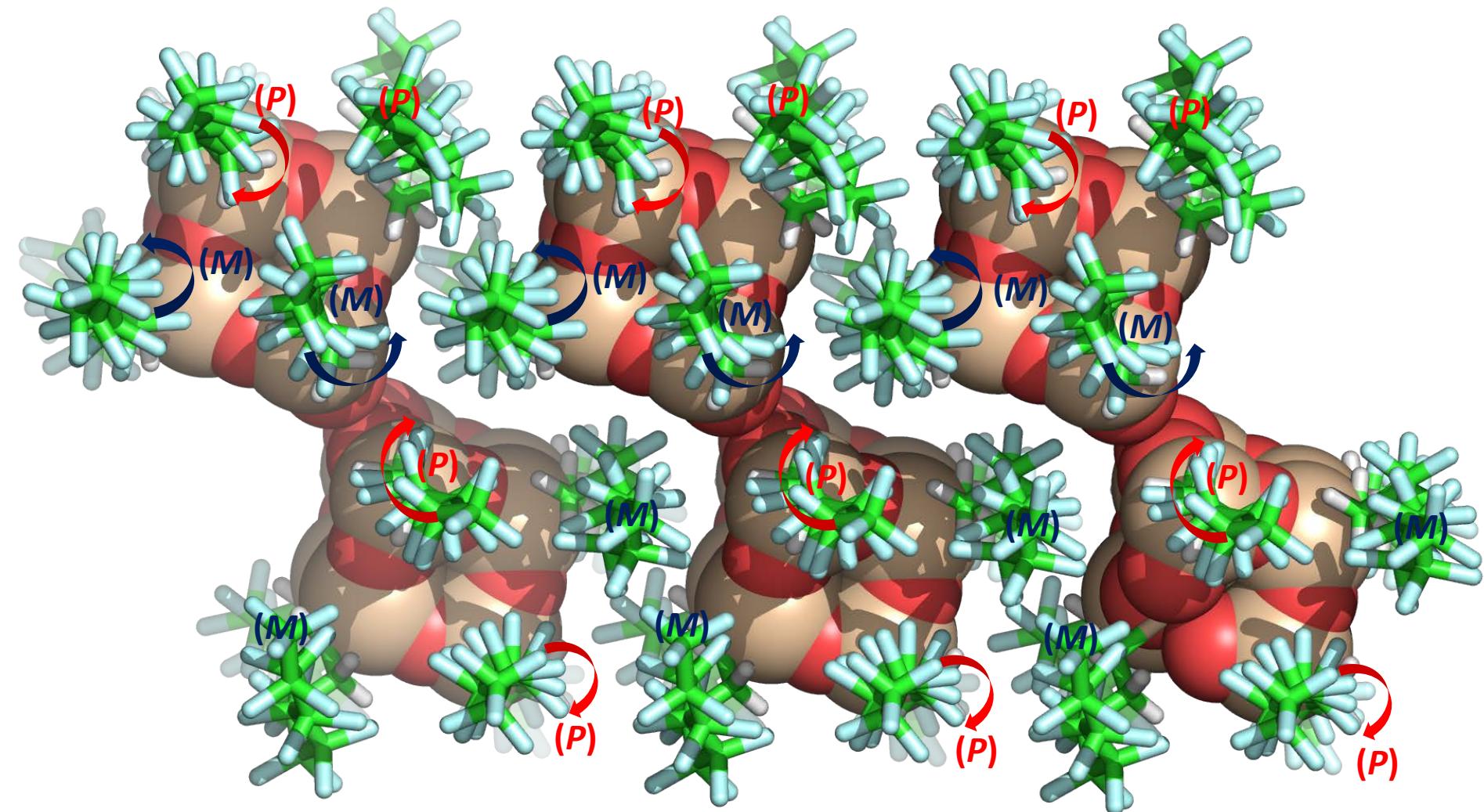
Clockwise (P)

A view from the top

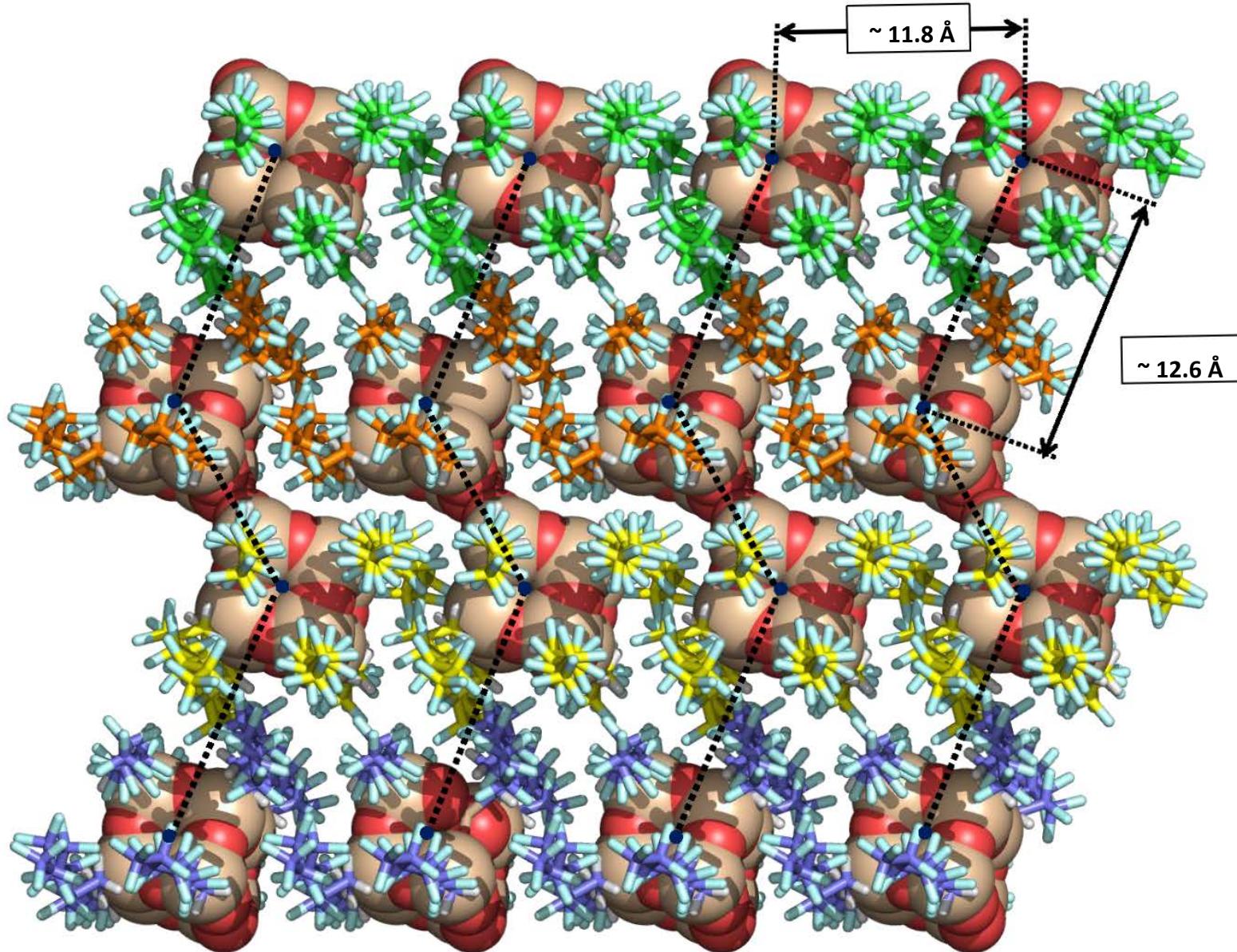
**F-POSS cage dimer (fluoroalkyl side chains shown as golden spheres):**  
**2 intermolecular OH...O - bonds that hold dimer structure together**



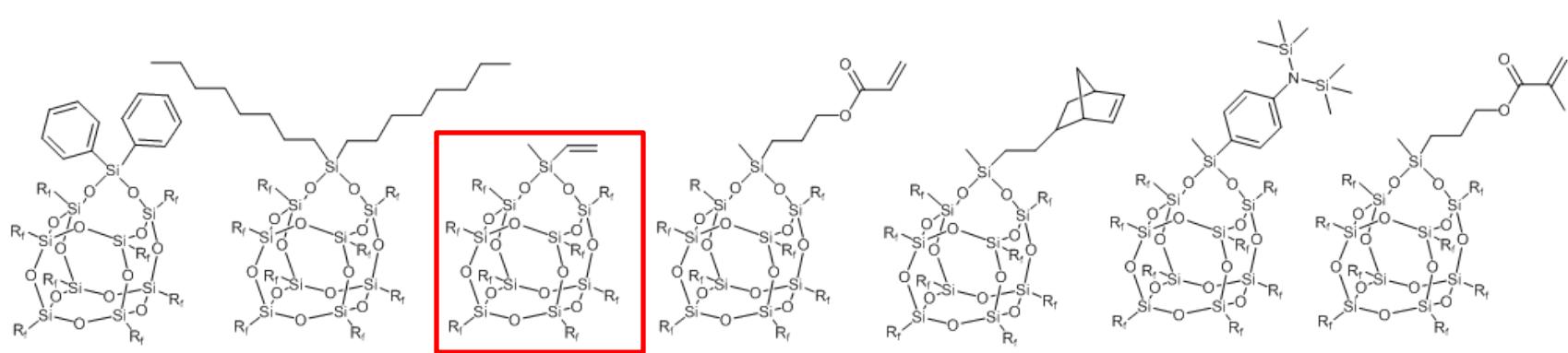
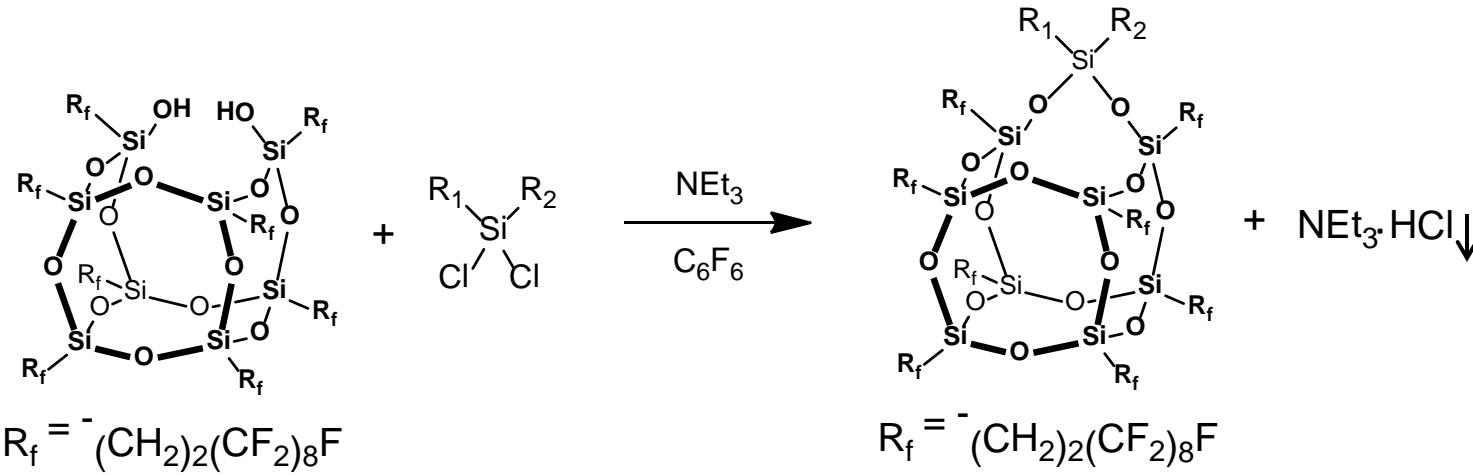
# F-POSS cage array: 3 dimers (6 cages) – top view



# F-POSS cage “zigzag” array (16 cages) – top view



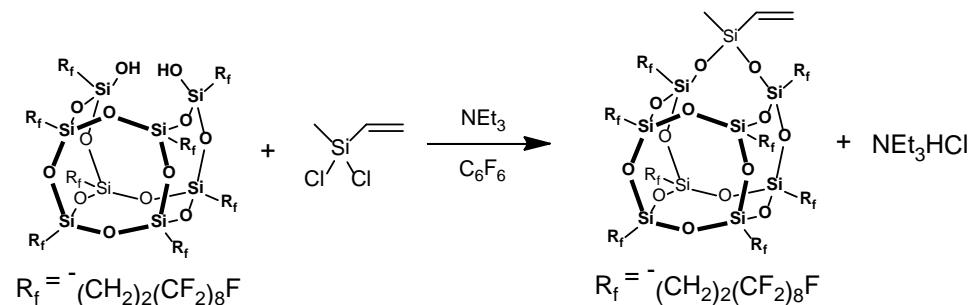
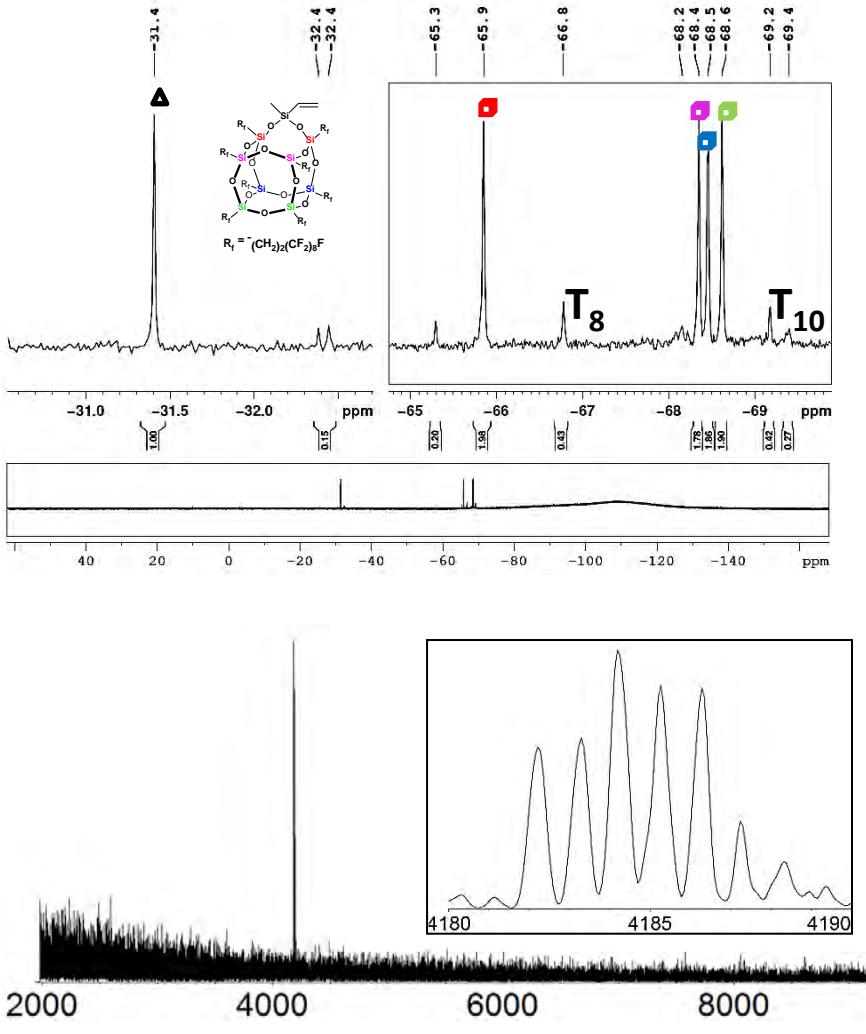
# Functional Fluorodecyl POSS Synthesis from Incompletely Condensed Fluorodecyl POSS



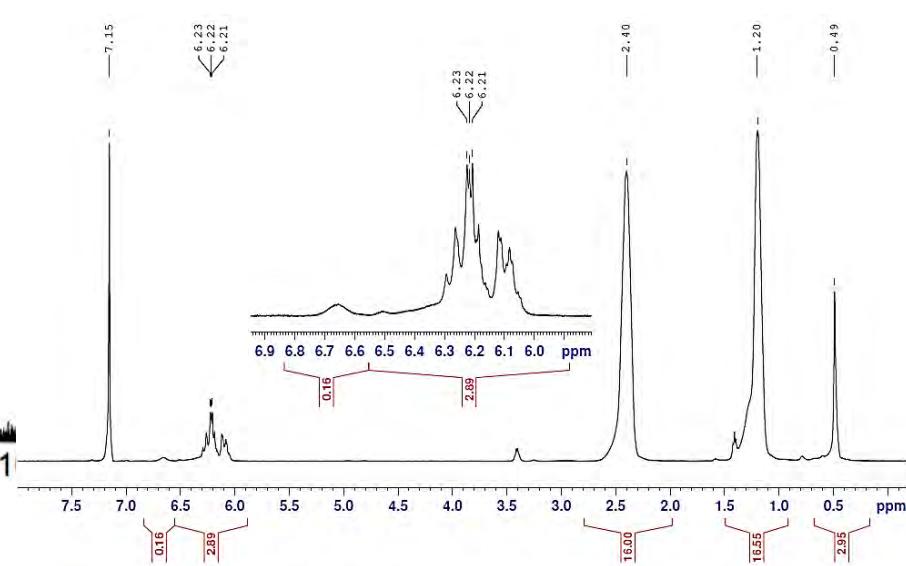
Ramirez, S. M., Diaz, Y. J., Campos, R., Stone, R. L., Haddad, T. S., Mabry, J. M. *JACS*, **2011**, *133*, 20084.

Distribution A: Approved for public release; distribution is unlimited.

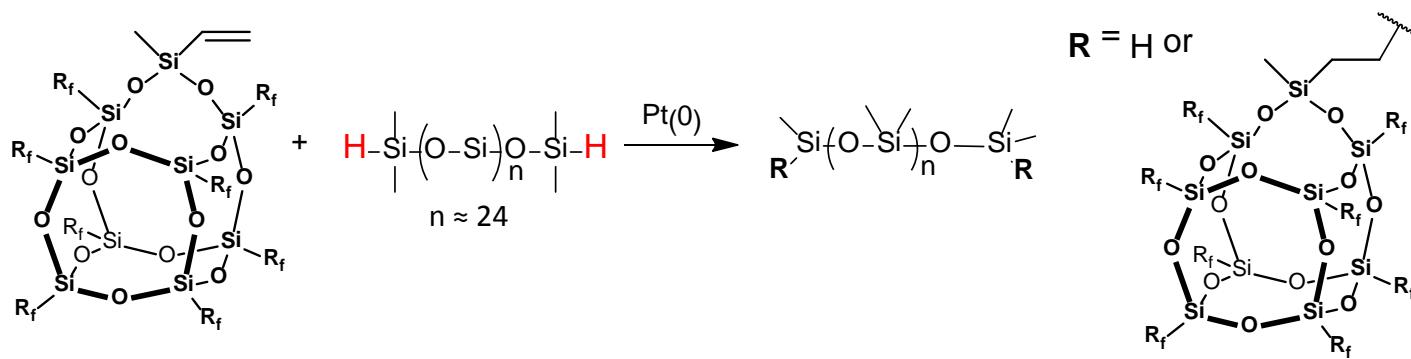
# Fluorodecyl<sub>8</sub>T<sub>8</sub>D<sub>1</sub>(methyl, vinyl) POSS



**Vinyl F-POSS**



# F-POSS Grafting to PDMS via Hydrosilylation

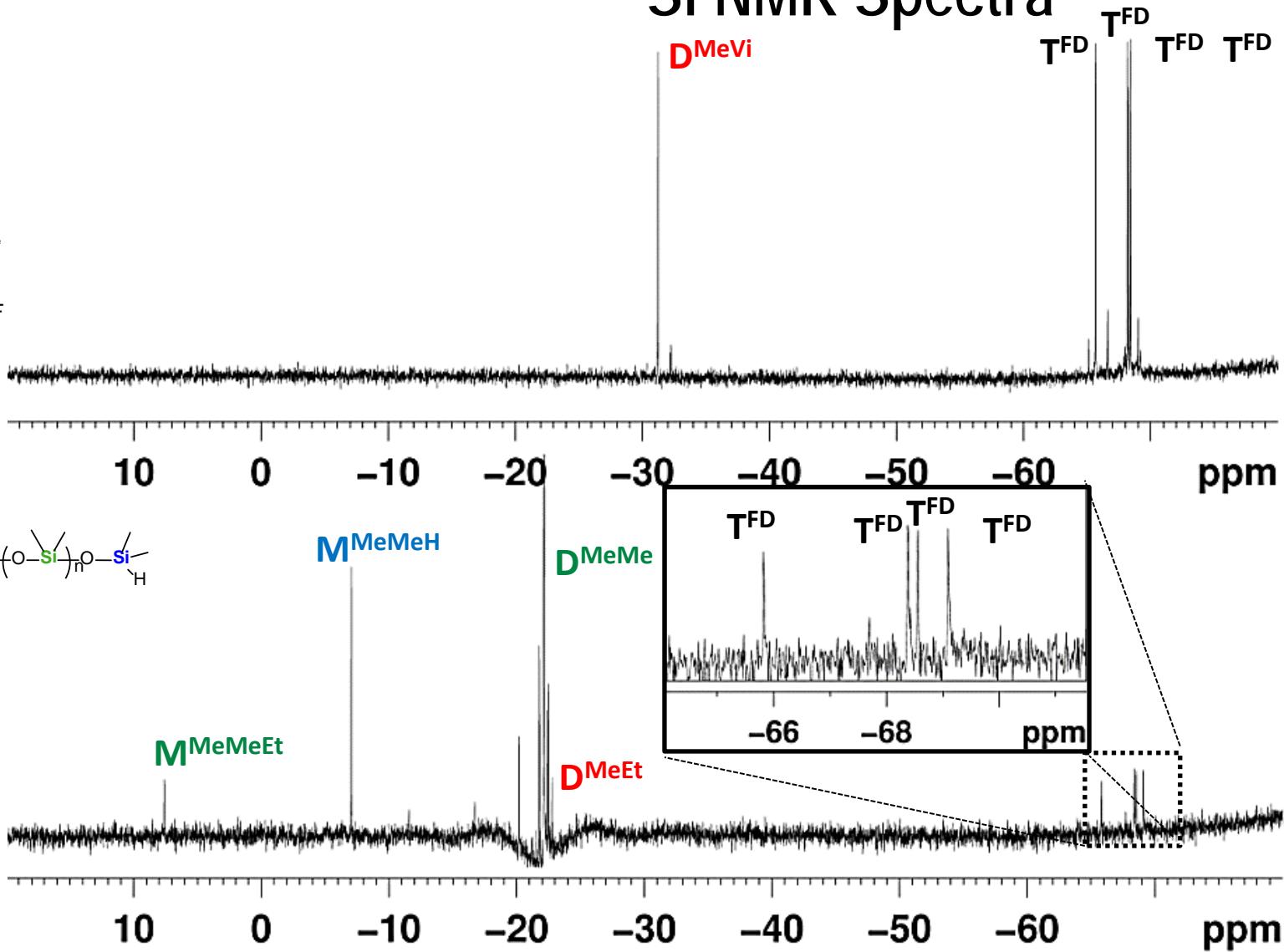
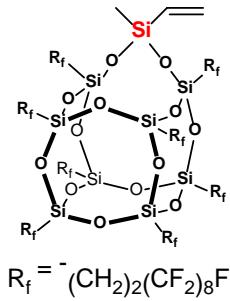


1, 5, 10, 20, 44 wt%

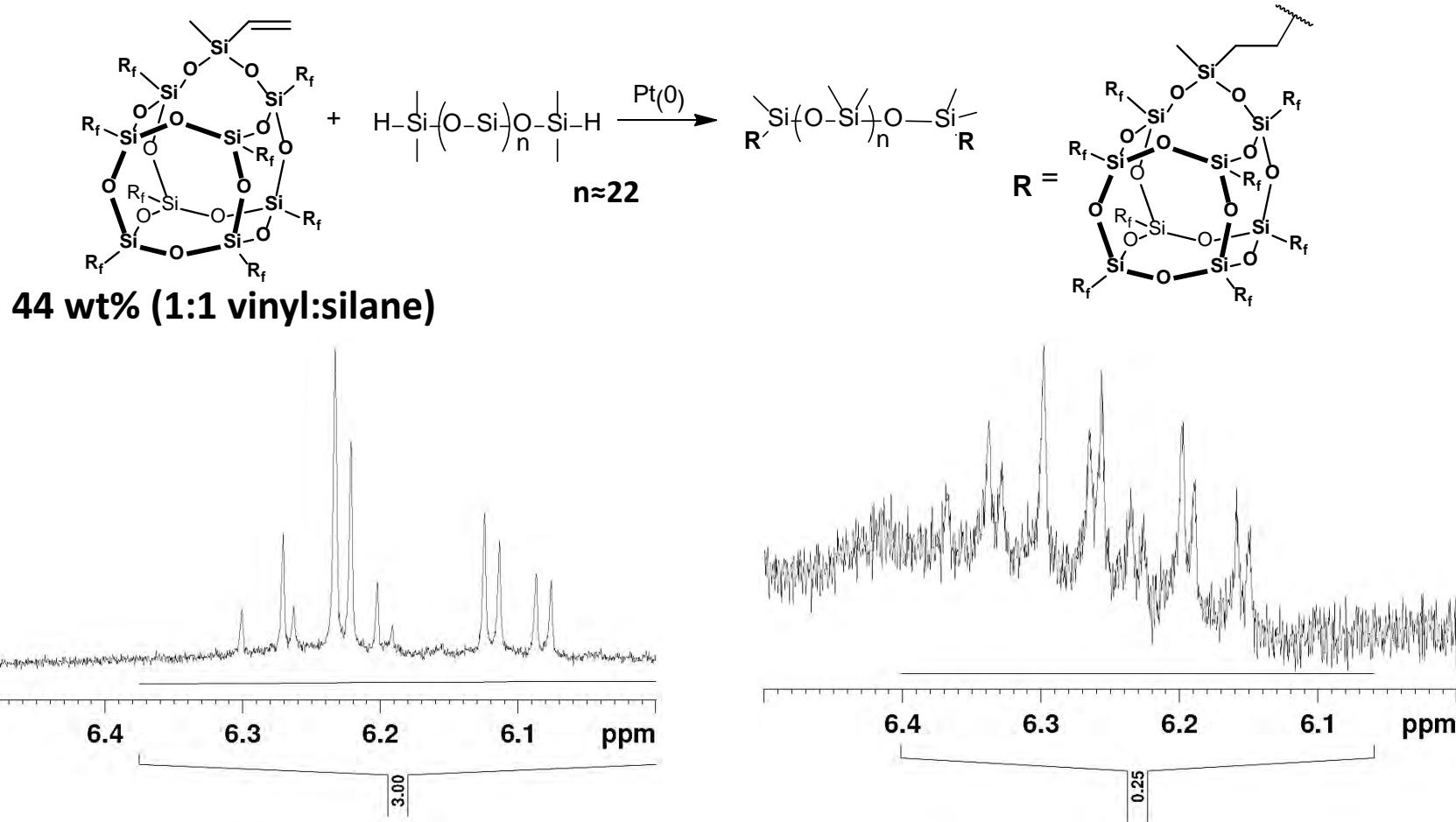
Vinyl: silane, 1:32, 1:8, 1:4, 1:2, 1:1

# Complete Conversion at 10 wt% F-POSS:

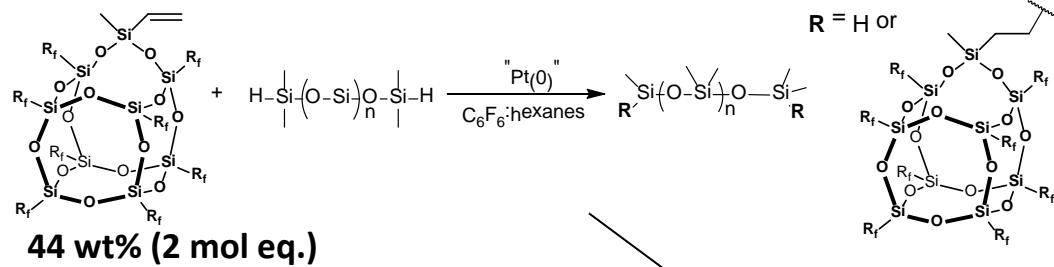
## $^{29}\text{Si}$ NMR Spectra



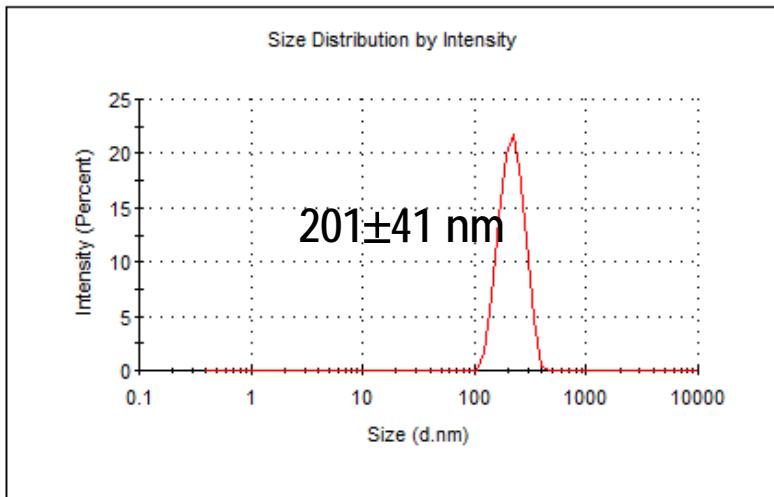
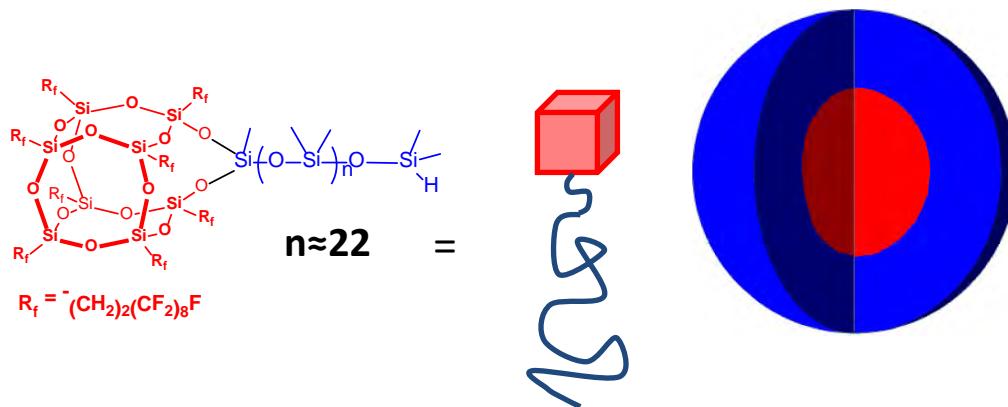
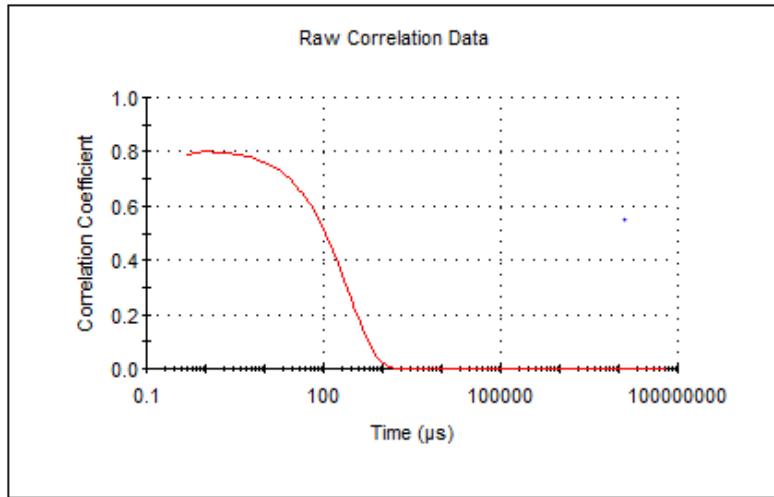
# Incomplete Conversion at $\geq$ 20 wt% F-POSS: $^1\text{H}$ NMR Spectra



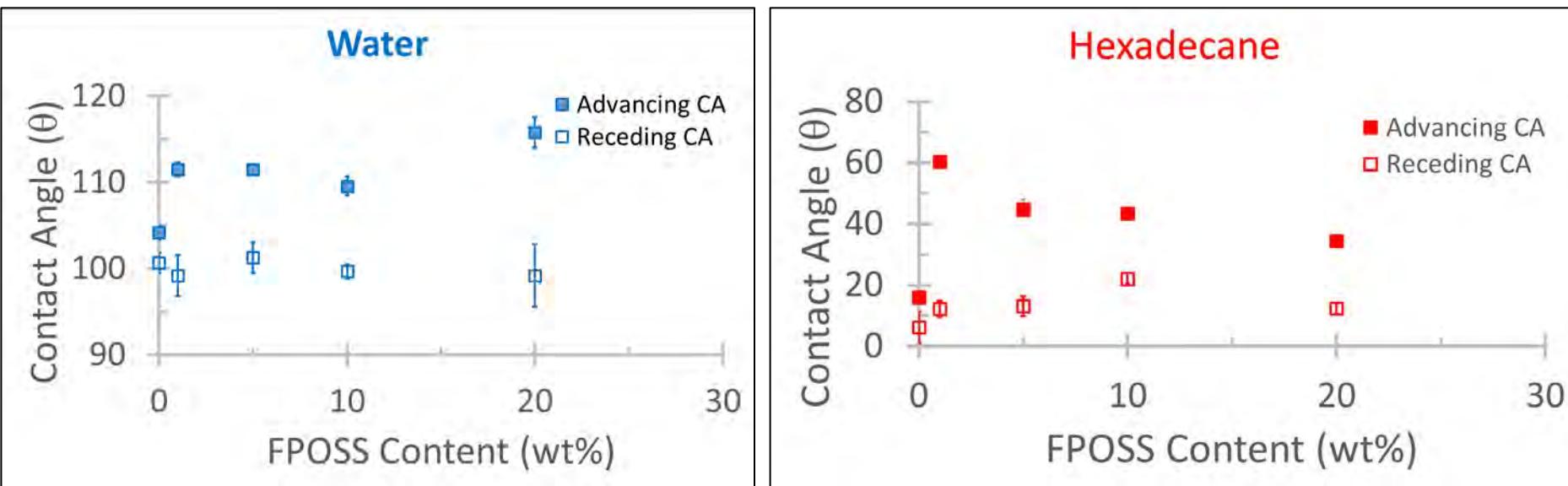
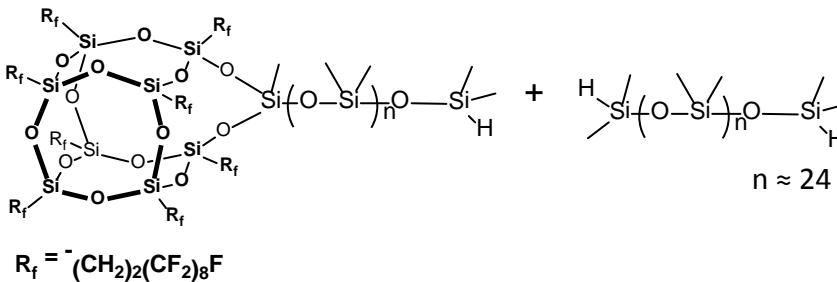
# F-POSS PDMS Amphiphile Aggregation



Crude rxn. mix.  
4:1  $C_6F_6$ :hexanes

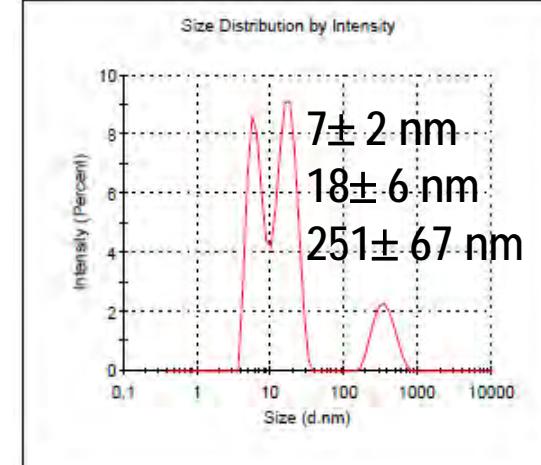
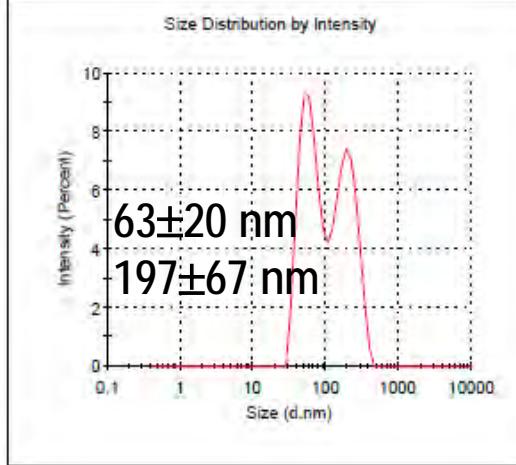
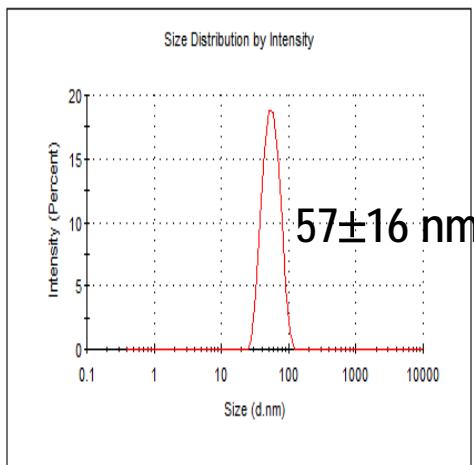
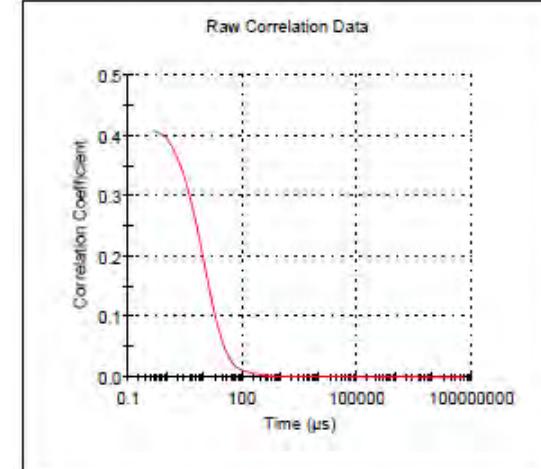
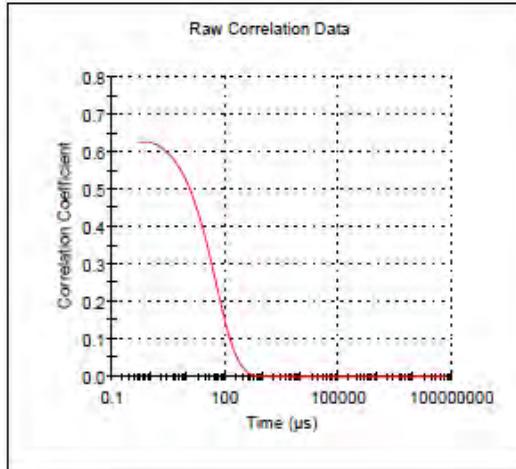
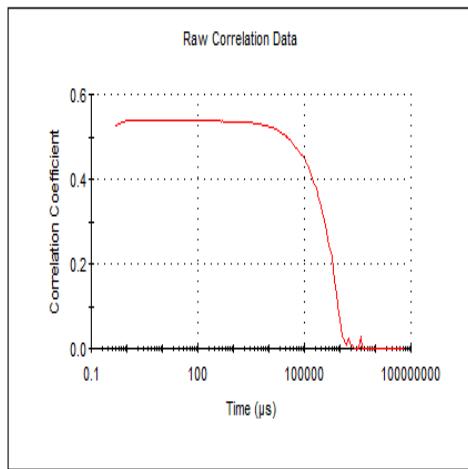


# Surface Wetting of Fluorodecyl POSS -PDMS Amphiphiles



Pure fluorodecyl<sub>8</sub>T<sub>8</sub> POSS: Water  $\theta_{adv}/\theta_{rec} = 124^\circ/116^\circ$   
Hexadecane  $\theta_{adv}/\theta_{rec} = 80^\circ/61^\circ$

# F-POSS PDMS Amphiphile Aggregation: Dynamic Light Scattering

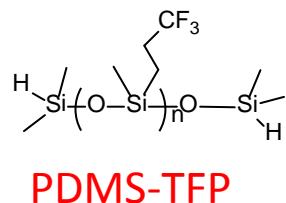
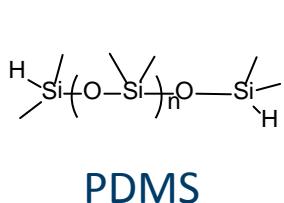


**PDMS (1000 cSt)**  
**0.001 M, 25 C**

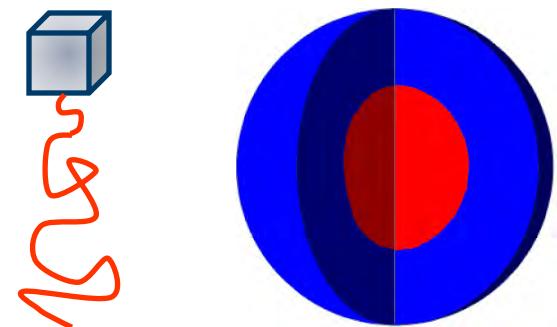
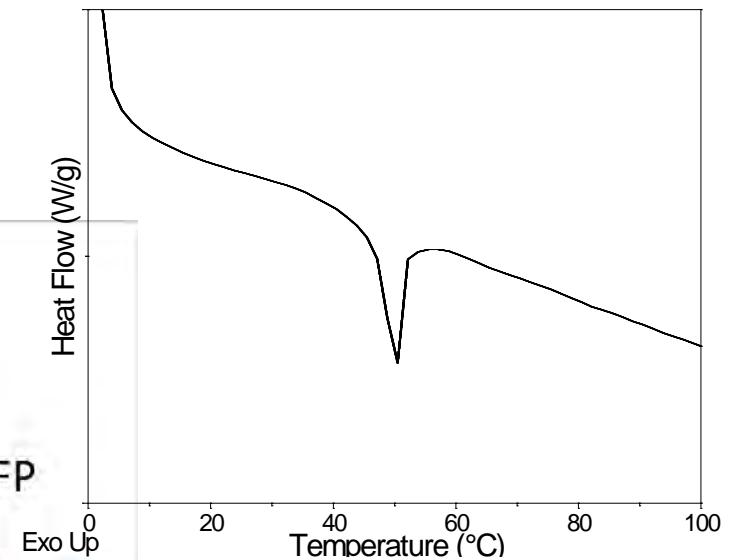
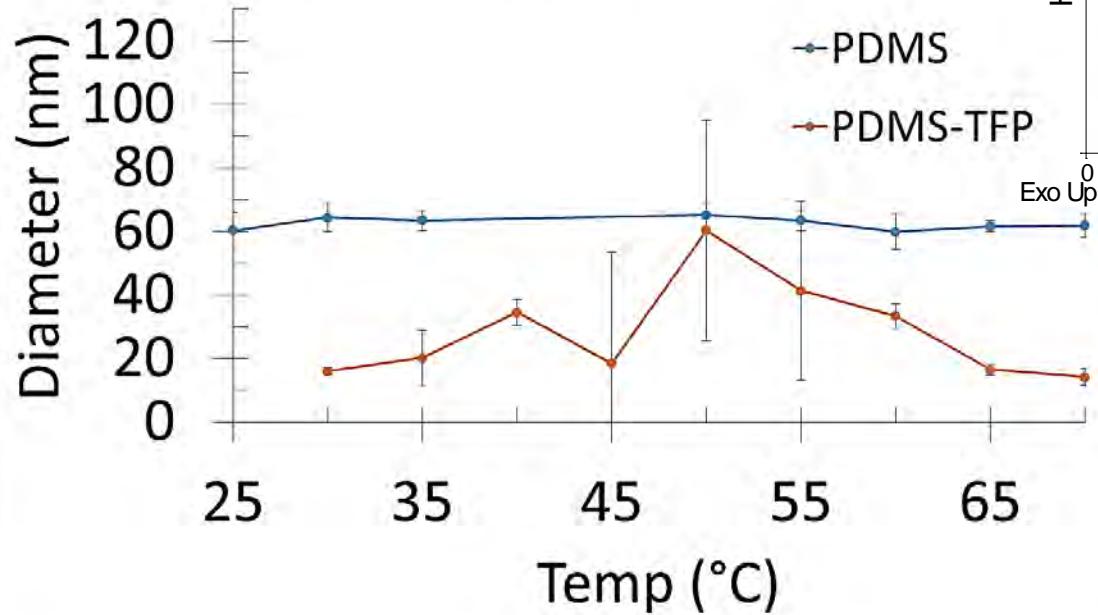
**CDCI3**  
**0.001 M, 25 C**

**AK225**  
**0.001 M, 25 C**

# Thermal Stability of FPOSS-PDMS Micelles

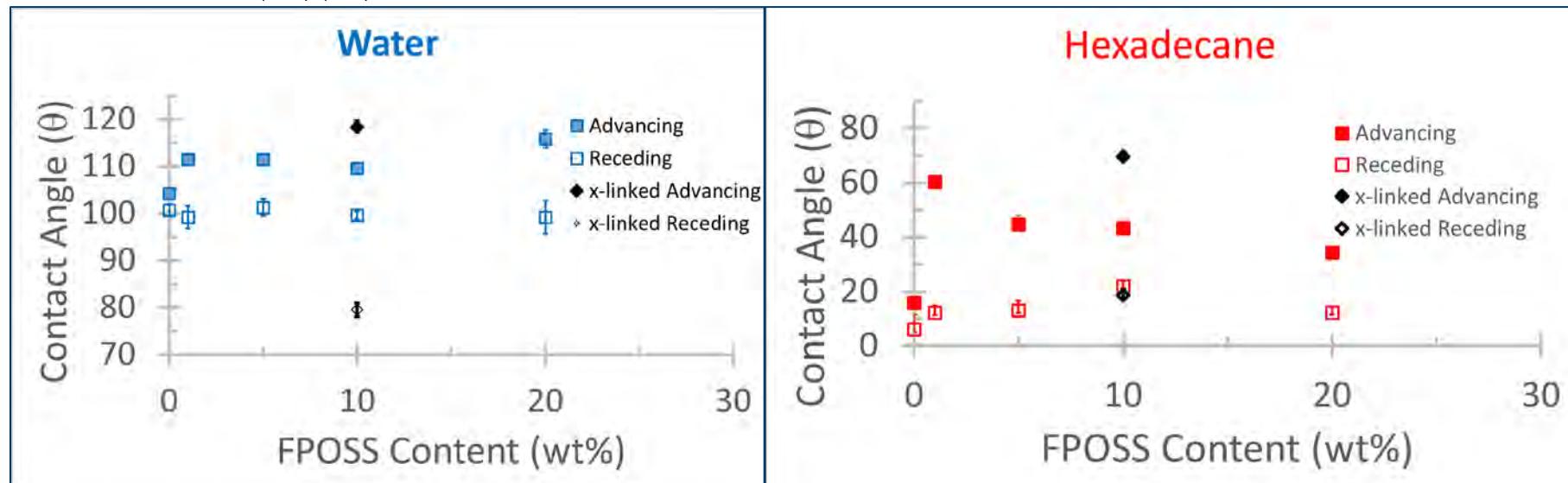
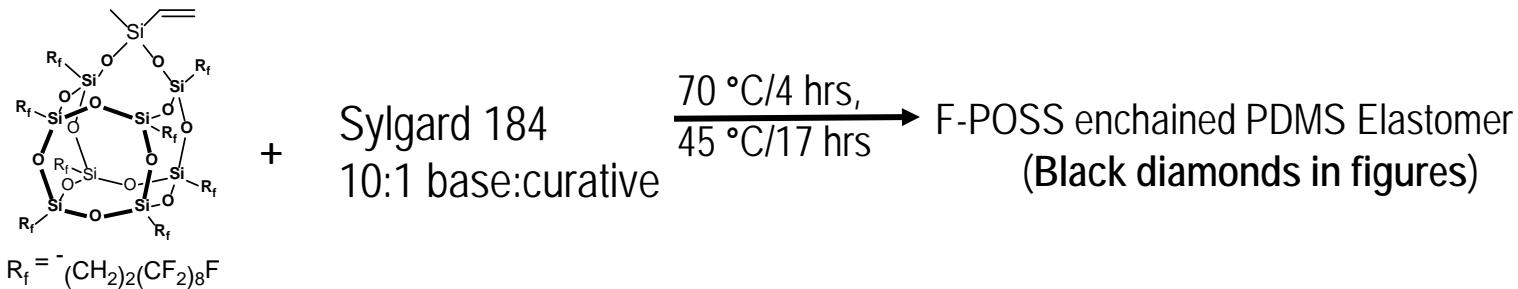


5 wt% FPOSS-PDMS



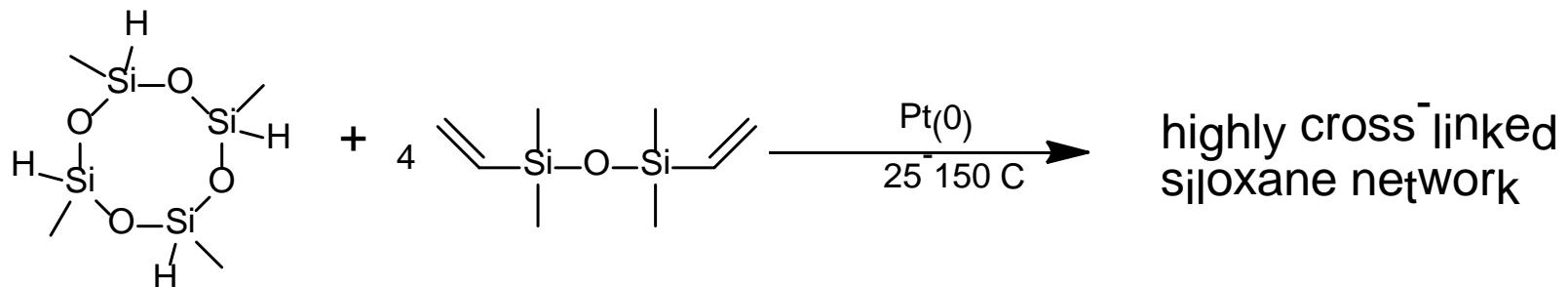
In fluorinated matrix

# Surface Wetting of Fluorodecyl POSS-Enchained PDMS Elastomers



Pure fluorodecyl<sub>8</sub>T<sub>8</sub> POSS: Water  $\theta_{adv}/\theta_{rec} = 124^\circ/116^\circ$   
Hexadecane  $\theta_{adv}/\theta_{rec} = 80^\circ/61^\circ$

# Surface modification of highly crosslinked siloxane networks



ethanol  $\Theta_{\text{stat}} / \Theta_{\text{adv}} / \Theta_{\text{rec}} = 8.9$

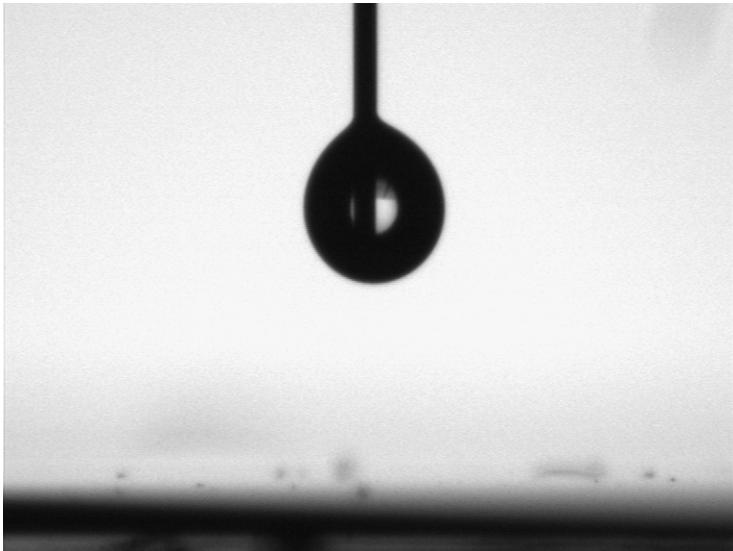
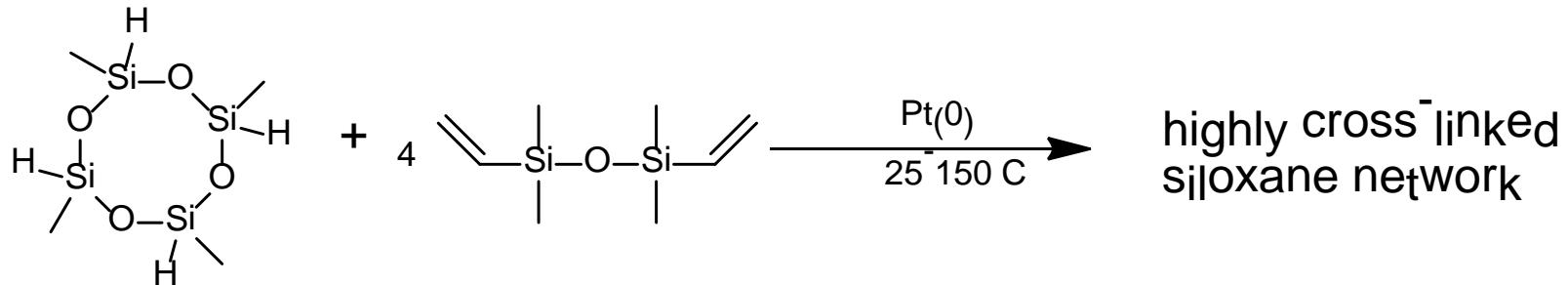
$22.88 \pm 2.46$   
0

isopropanol completely wets and spreads

$\theta_{\text{adv}} / \theta_{\text{rec}} (\circ)$

Water	$99.5 \pm 0.6 / 93.2 \pm 0.5$
Hexadecane	$33.6 \pm 0.3 / 28.9 \pm 0.3$
Heptane	$<5 / 0$
methanol	$29.3 \pm 0.5 / 18.3 \pm 2.4$

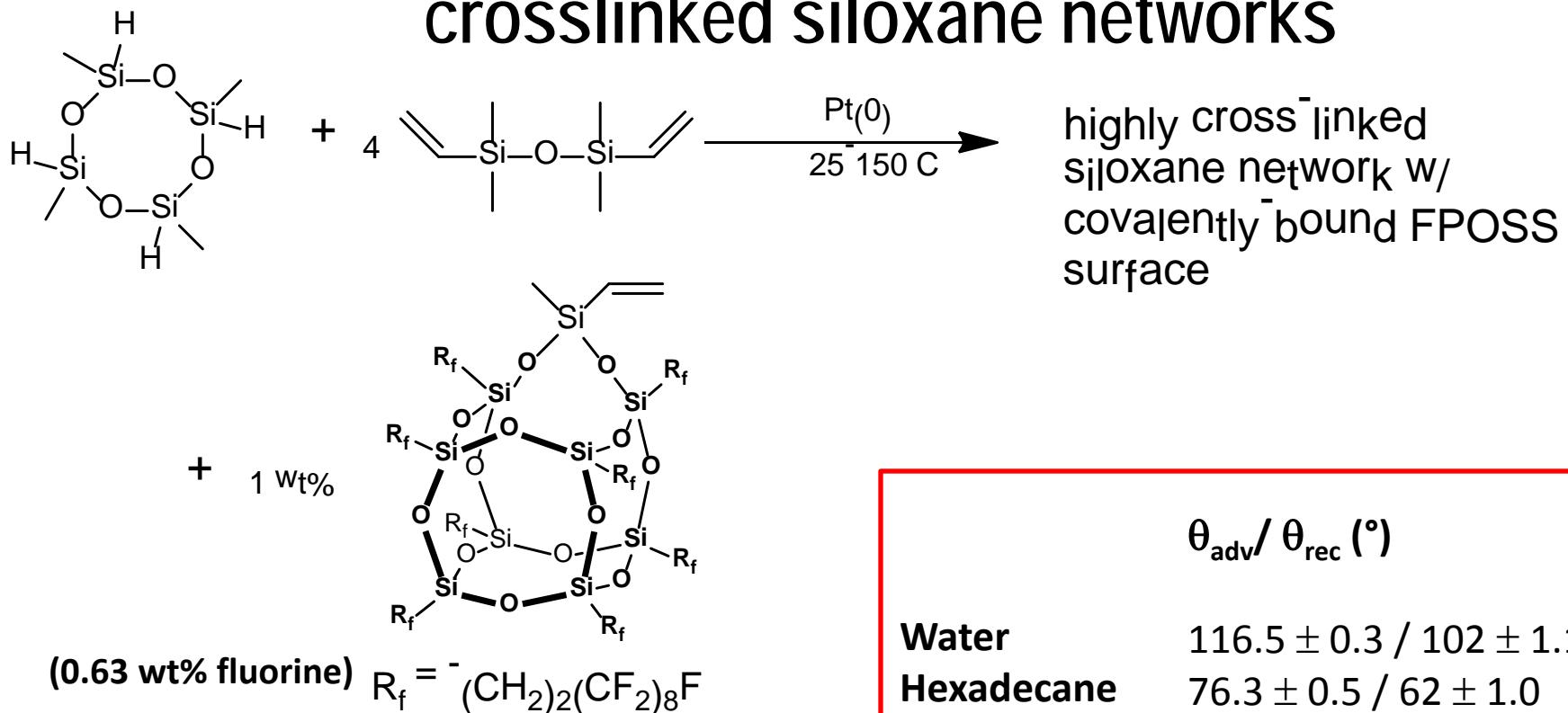
# Surface modification of highly crosslinked siloxane networks



$\theta_{\text{adv}} / \theta_{\text{rec}} (\circ)$

<b>Water</b>	$99.5 \pm 0.6 / 93.2 \pm 0.5$
<b>Hexadecane</b>	$33.6 \pm 0.3 / 28.9 \pm 0.3$
<b>Heptane</b>	$<5 / 0$
<b>methanol</b>	$29.3 \pm 0.5 / 18.3 \pm 2.4$

# Surface modification of highly crosslinked siloxane networks

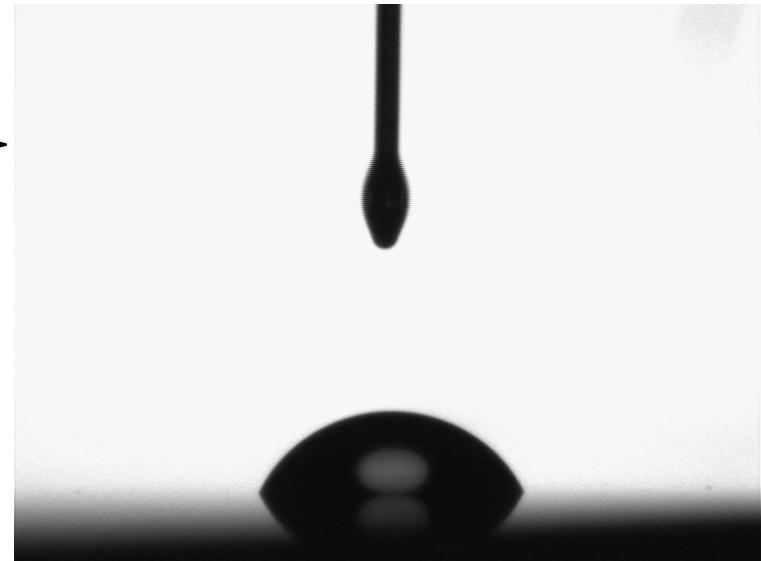
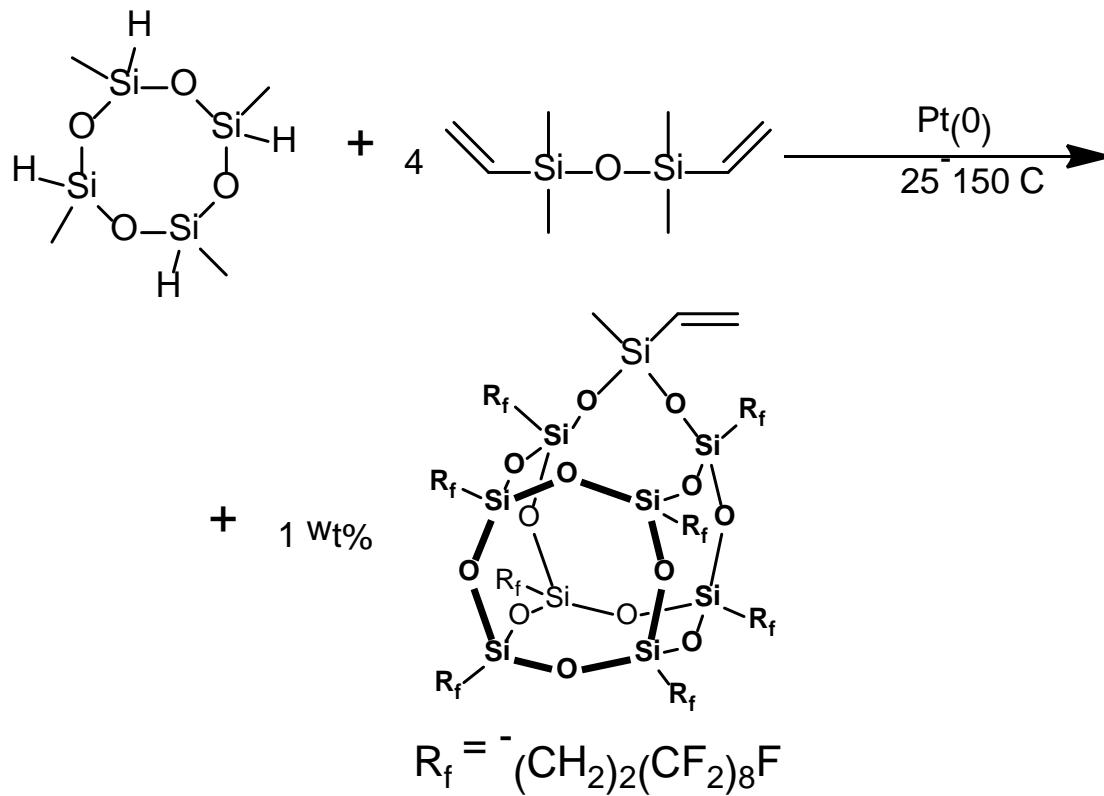


isopropanol  $\Theta_{\text{stat}} / \Theta_{\text{adv}} / \Theta_{\text{rec}} = 54.35 \pm 4.41$   
 $70.94 \pm 1.05$   
 $40.5 \pm 0.86$

ethanol  $\Theta_{\text{stat}} / \Theta_{\text{adv}} / \Theta_{\text{rec}} = 60.588 \pm 7.62$   
 $73.97 \pm 1.0075$   
 $41.4 \pm 1.2725$

	$\theta_{\text{adv}} / \theta_{\text{rec}} (\circ)$
Water	$116.5 \pm 0.3 / 102 \pm 1.1$
Hexadecane	$76.3 \pm 0.5 / 62 \pm 1.0$
Heptane	$56.1 \pm 0.9 / 39.2 \pm 1.2$
methanol	$75.2 \pm 0.9 / 49.9 \pm 1.5$

# Surface modification of highly crosslinked siloxane networks



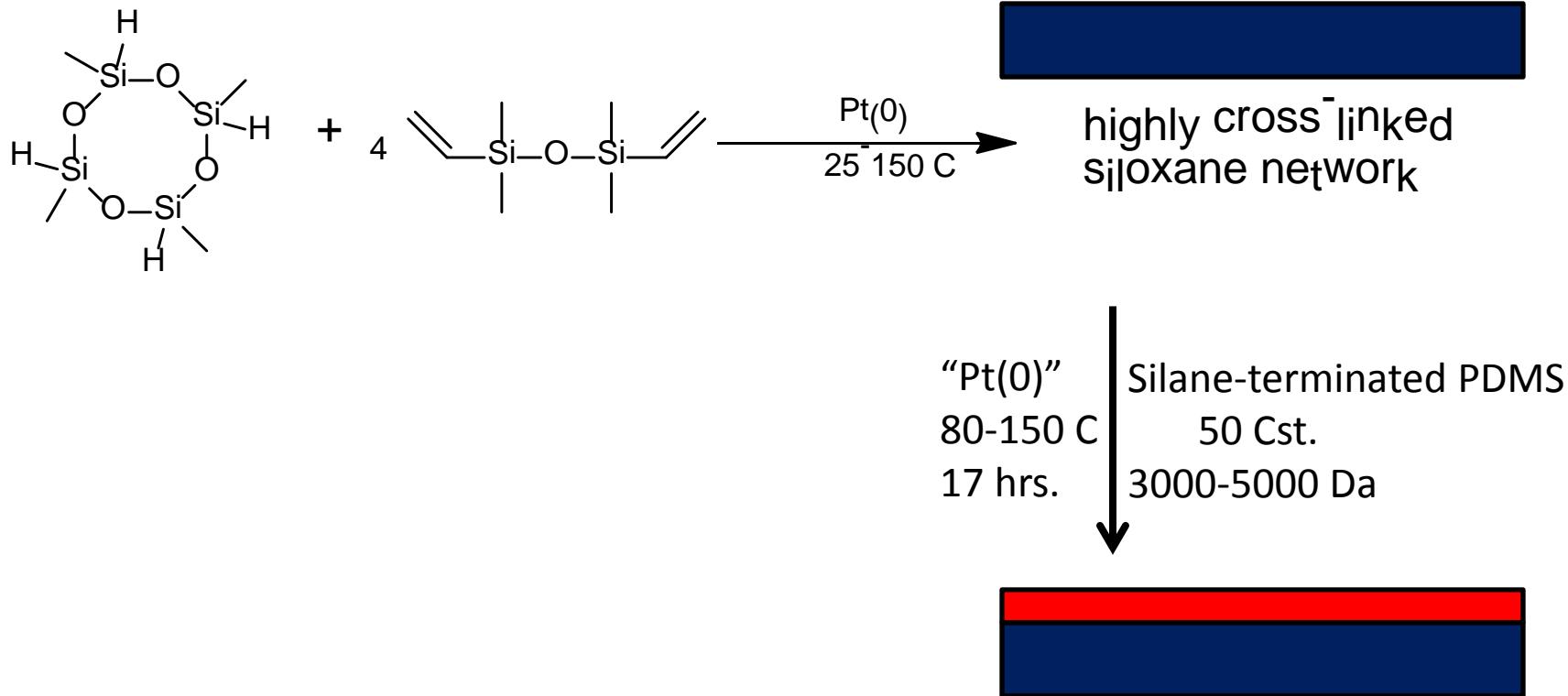
$\theta_{adv} / \theta_{rec}$  (°)

Water	$116.5 \pm 0.3 / 102 \pm 1.1$
Hexadecane	$76.3 \pm 0.5 / 62 \pm 1.0$
Heptane	$56.1 \pm 0.9 / 39.2 \pm 1.2$
methanol	$75.2 \pm 0.9 / 49.9 \pm 1.5$

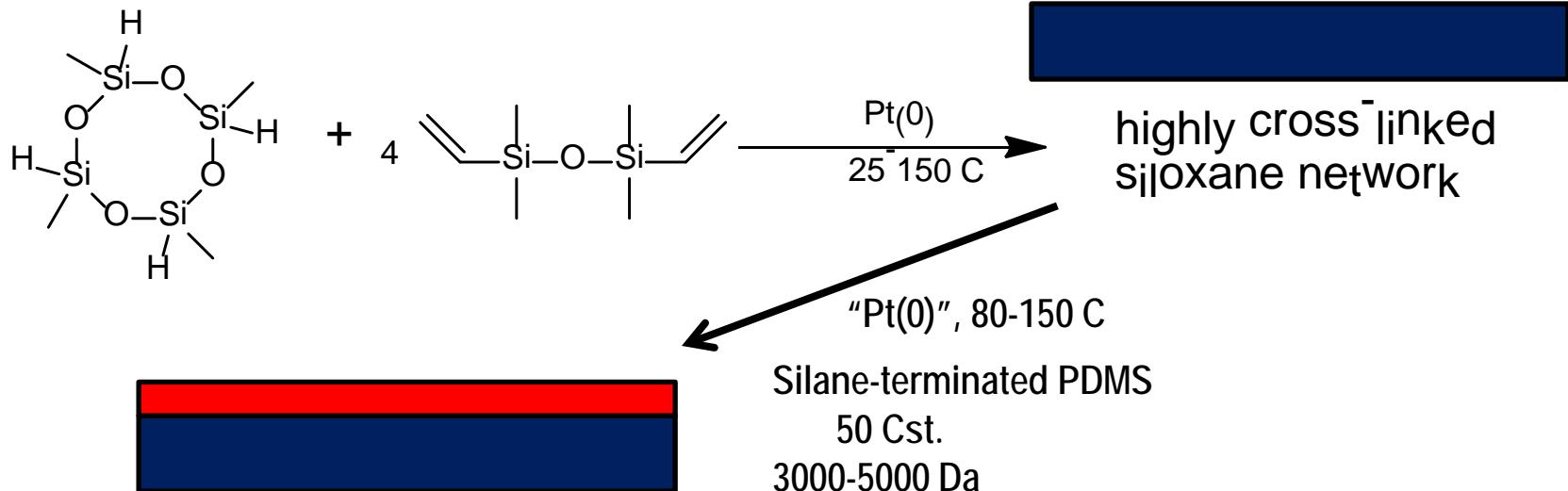
# What about “liquid-like” surfaces and post-cure surface modification?

.....

# Post-cure surface modification of highly crosslinked siloxane networks with silane-term oligo(dimethylsiloxane)



# Post-cure surface modification of highly crosslinked siloxane networks with silane-term oligo(dimethylsiloxane)



$\theta_{\text{adv}} / \theta_{\text{rec}}$  ( $^\circ$ )

Water	$110.7 \pm 0.3 / 89.4 \pm 0.5$
Hexadecane	$47.8 \pm 0.4 / 17.5 \pm 0.7$
Heptane	$7.3 \pm 1.9 / 0$
methanol	$40.8 \pm 0.4 / 35.5 \pm 0.6$

$\theta_{\text{adv}} / \theta_{\text{rec}}$  ( $^\circ$ )

Water	$99.5 \pm 0.6 / 93.2 \pm 0.5$
Hexadecane	$33.6 \pm 0.3 / 28.9 \pm 0.3$
Heptane	$<5 / 0$
methanol	$29.3 \pm 0.5 / 18.3 \pm 2.4$